



UNIVERSITEIT VAN AMSTERDAM



From Batch to Flow: Advancing Synthetic Organic Chemistry through Technological Innovation

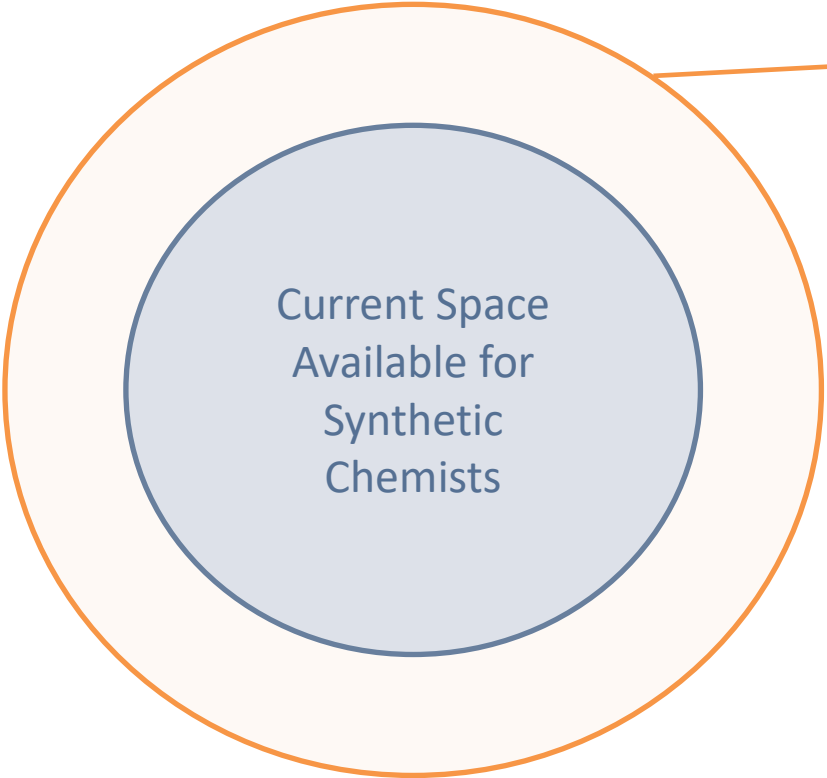
Prof. Dr. ing. Timothy Noël

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University of Amsterdam
The Netherlands

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 @NoelGroupUvA, @tnoel82

Reimagining Synthetic Chemistry



Current Space
Available for
Synthetic
Chemists

Our mission:

Expanding the available chemical space by embracing technology to the fullest extent

by

- 1) *Developing new synthetic transformations using reagents or conditions that are difficult to handle*
- 2) *Developing new tools to make synthesis easier*
- 3) *Showing unique selectivity and reactivity*

How?

*by merging
organic chemistry and chemical engineering*

How can flow make an impact?

Inherent advantages of microscale flow reactors:

Enhanced mass transfer

Enhanced heat transfer

High reproducibility

Multistep flow sequences

Automation

Safety

Scalability

Gas-liquid reactions

Taming exothermic reactions

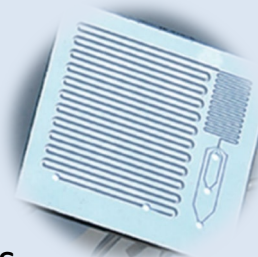
Reaction kinetics

Time-gain, labor reduction

Minimizing human error

New processing windows

From mg to kg in same device



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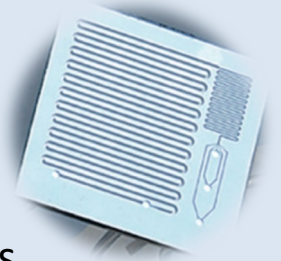
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Problem: In general, people resist change.

Education is Key!

OPINION

Flow into the chemistry curriculum

BY TIMOTHY NOEL | 27 SEPTEMBER 2019

There's more to chemistry than the round-bottomed flask

It's ironic that chemists are experts at change, except when it comes to their own practice. Mark Gilligan recently wrote about chemists' reluctance to adopt flow chemistry as an example of this innate resistance to change. I have seen that same resistance, and I understand it. Why would you suddenly change your habits and embrace an expensive new technology?



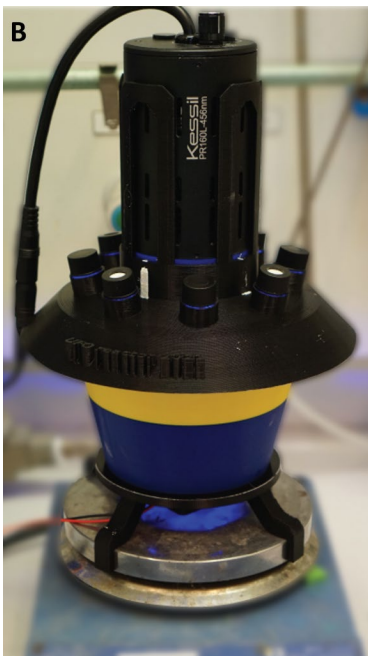
“Put flow chemistry in your curriculum and give students the broadest experience of making molecules. Let them decide which ideas have a future.”

For an opinion article: Noël, *Chemistry World* 2019, <https://www.chemistryworld.com/opinion/flow-into-the-chemistry-curriculum/4010382.article> .

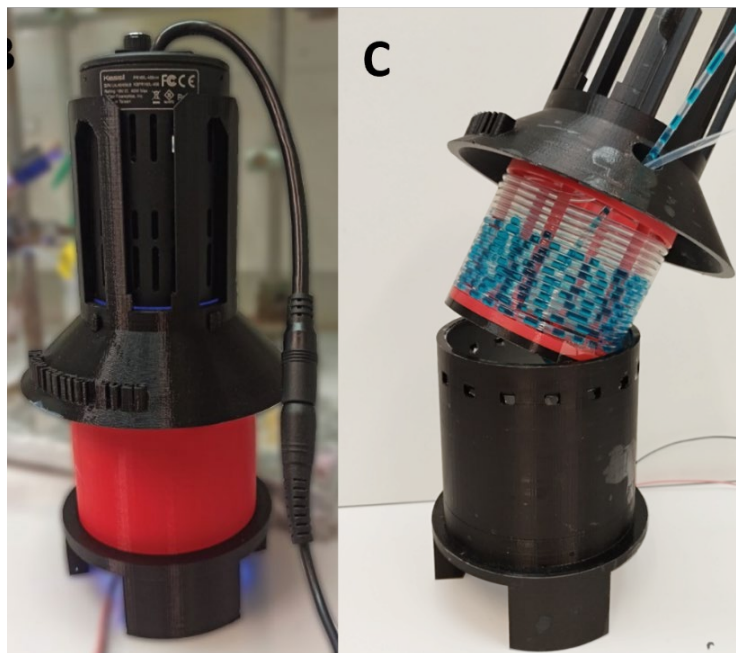
For our undergrad flow experiments: Kuijpers, Weggemans, Verwijlen, Noël, *J. Flow Chem.* 2021, 11, 7-12.

Democratization of Flow Chemistry

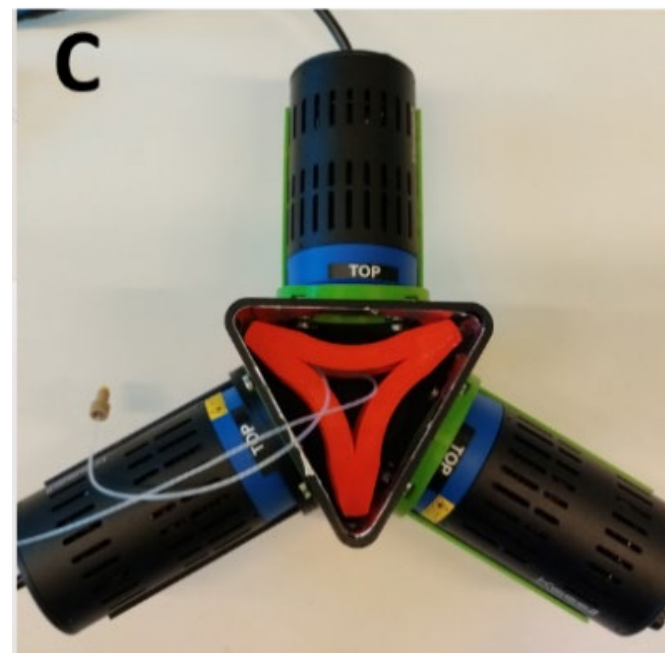
UFO – Batch



UFlow



Flow – Fidget Reactor



Fully characterized, standardized batch and flow setups that are affordable.

Designs available via: Masson, Zondag, Schuurmans, Noël, *React. Chem. Eng.* **2024**, 9, 2218–2225.

For characterization procedure: Zondag, Schuurmans, Chaudhuri, Visser, Soares, Padoin, Kuijpers, Dorbec, van der Schaaf, Noël, *Nature Chemical Engineering* **2024**, 1, 462–471.

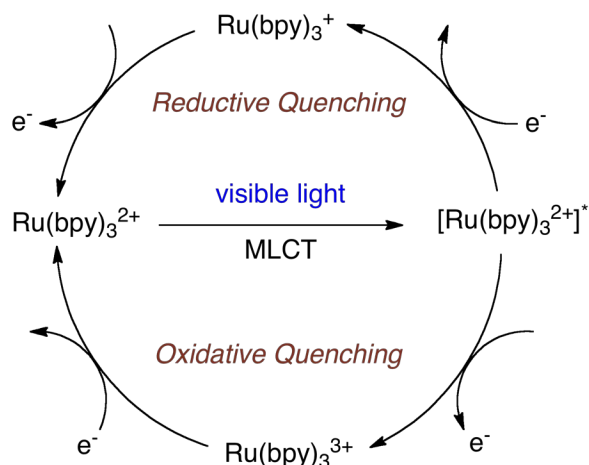
Photocatalysis

Photoredox Catalysis

photocatalysis allows for absorption of wavelengths of the UV-A and VIS

- Advantages:
- cheap, energy-efficient and high intensity energy light sources (LEDs)
 - mild reaction conditions (room temperature, functional group tolerance)
 - new opportunities in organic synthesis

$\text{Ru}(\text{bpy})_3\text{Cl}_2$ is the most studied one-electron photoredox catalyst.



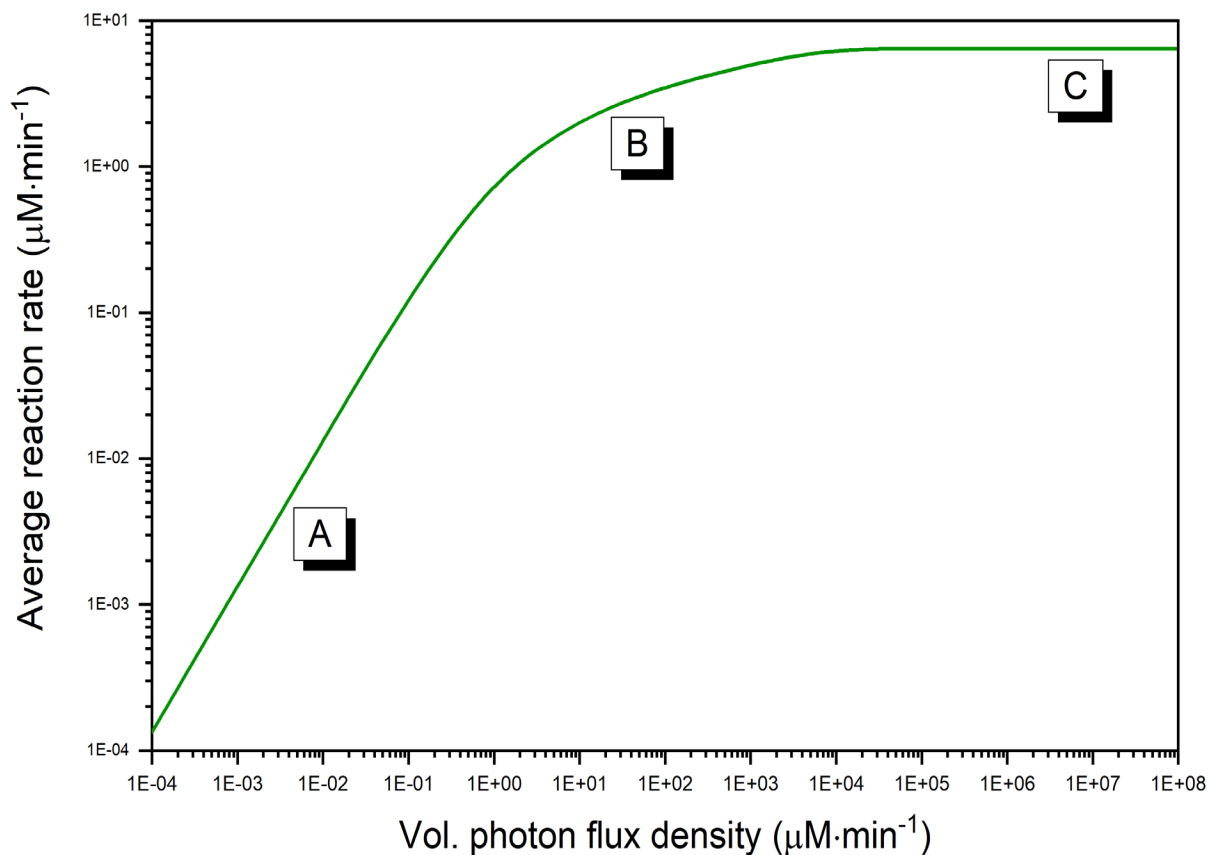
Special issue in *Chemical Reviews* on **Photochemical Catalytic Processes** (Paolo Melchiorre, Guest Editor), **2022**, *122*, 1483-2980.

For a perspective on PC: Noel, Zysman-Colman, *Chem Catalysis* **2022**, *2*, 468-476.

Light activation of molecules

Thermochemical activation: $k = A \left(-\frac{E_a}{RT} \right)$

Photochemical activation: $k = \alpha \cdot I^\beta$



For comprehensive reviews: (i) Cambié, Bottecchia, Straathof, Hessel, Noël, *Chem. Rev.* **2016**, *116*, 10276-10341. (ii) Buglioni, Raymenants, Slattery, Zondag, Noël, *Chem. Rev.* **2022**, *122*, 2752-2906.

Industrial Importance of Photoredox Catalysis

*Every single pharma- and agro-chemical company
has initiated programs to implement Photoredox catalysis.*

Industrial Importance of Photoredox Catalysis

Every single pharma- and agro-chemical company has initiated programs to implement Photoredox catalysis.

Medicinal Chemistry:

- Goal: identify new chemical structures ASAP.
- Small amounts for bio-assays and ADME studies
- Well implemented

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Process Chemistry:

- Goal: clean, cost-effective manufacturing process for new medicines
- Scalable process for clinical trials and commercialization
- Challenging!

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Scale up: 50 kg product/day is a good estimate for what is required in pharma!

This requires about 1000 W of optical power per day.

Industrial Importance of Photoredox Catalysis

1000 W of optical power per day to produce 50kg/day

To put it in perspective:

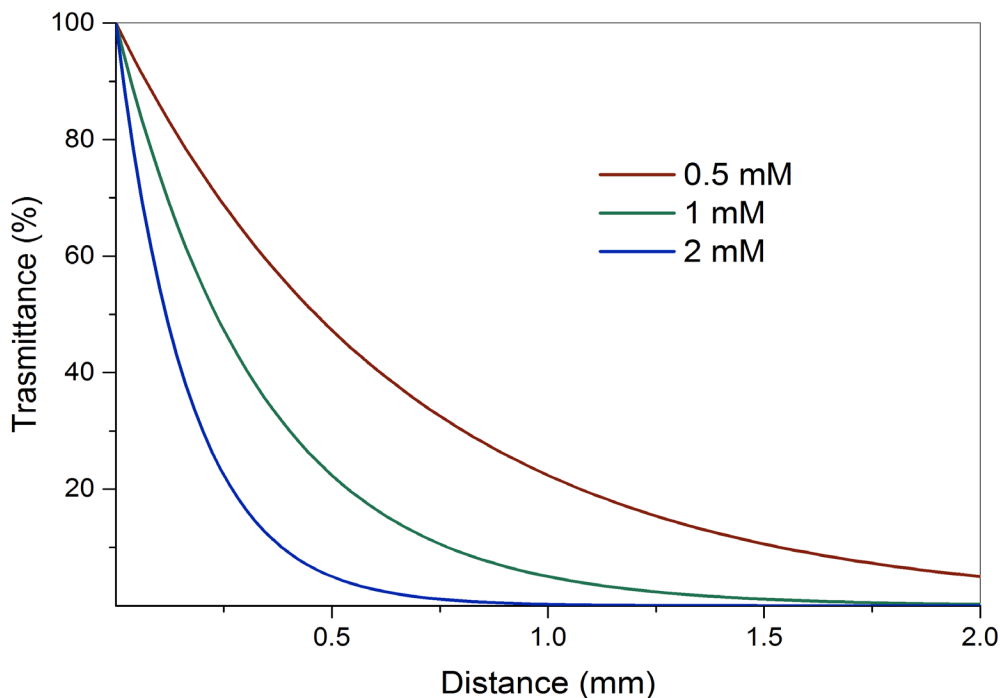
This amounts to the light delivered by 5000 CFL light bulbs !!!



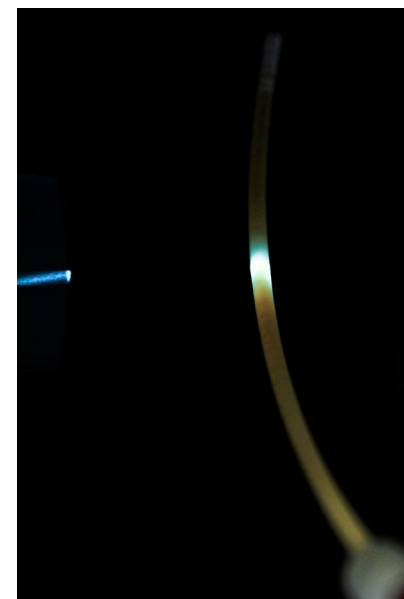
Photocatalysis scale up problems

Batch : limited penetration depth of irradiation because of absorption results in longer reaction times, higher catalyst loadings and difficult scale-up

$$\log(T) = \log(I_0/I) = \epsilon \cdot l \cdot c \quad (\text{Bouguer-Lambert-Beer})$$



Batch

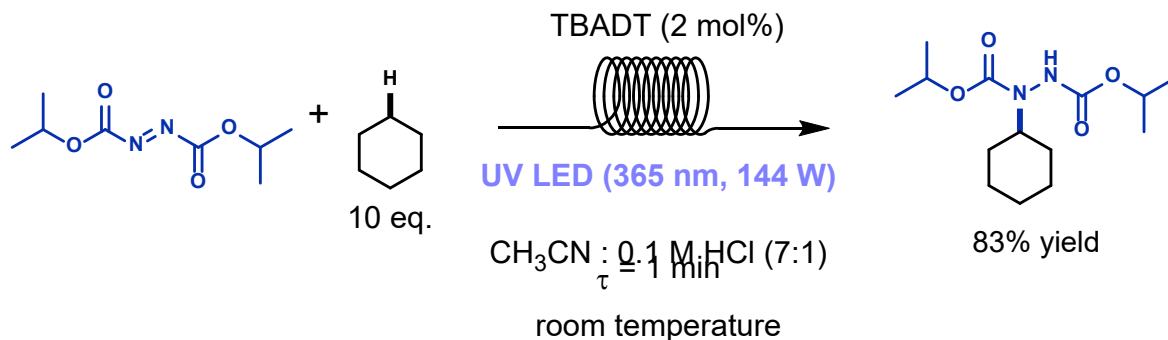


Flow

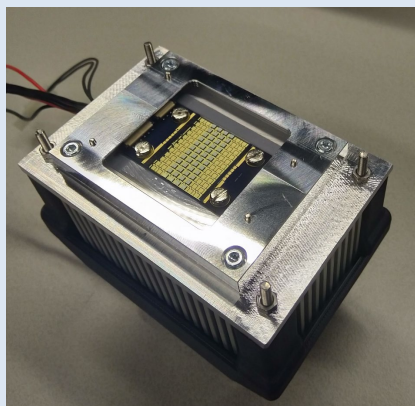
For comprehensive reviews: (i) Cambié, Bottecchia, Straathof, Hessel, Noël, *Chem. Rev.* **2016**, *116*, 10276-10341. (ii) Buglioni, Raymenants, Slattery, Zondag, Noël, *Chem. Rev.* **2022**, *122*, 2752-2906.

Scaling Photochemistry

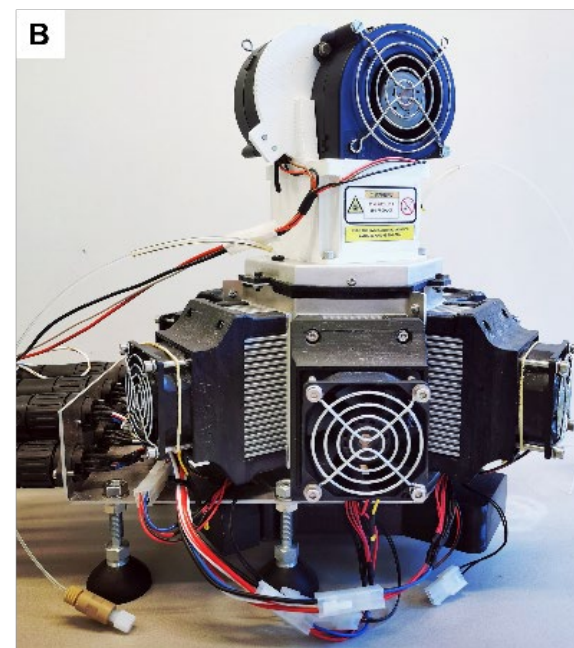
Longer operation times + intensified reaction conditions:



Reactor : 11 mL volume, ID 750 μm
Productivity : 314 mol/h (2.15 kg/day)

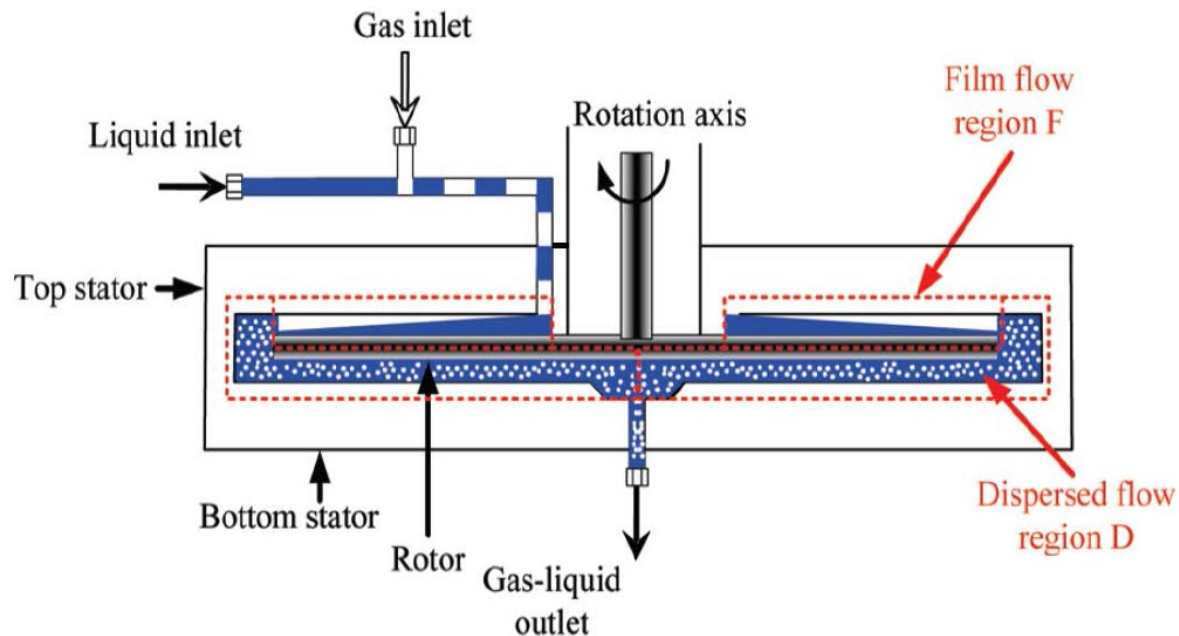


Chip-on-Board LEDs
(24 W optical power/unit)



Rotor-Stator Spinning Disk Technology

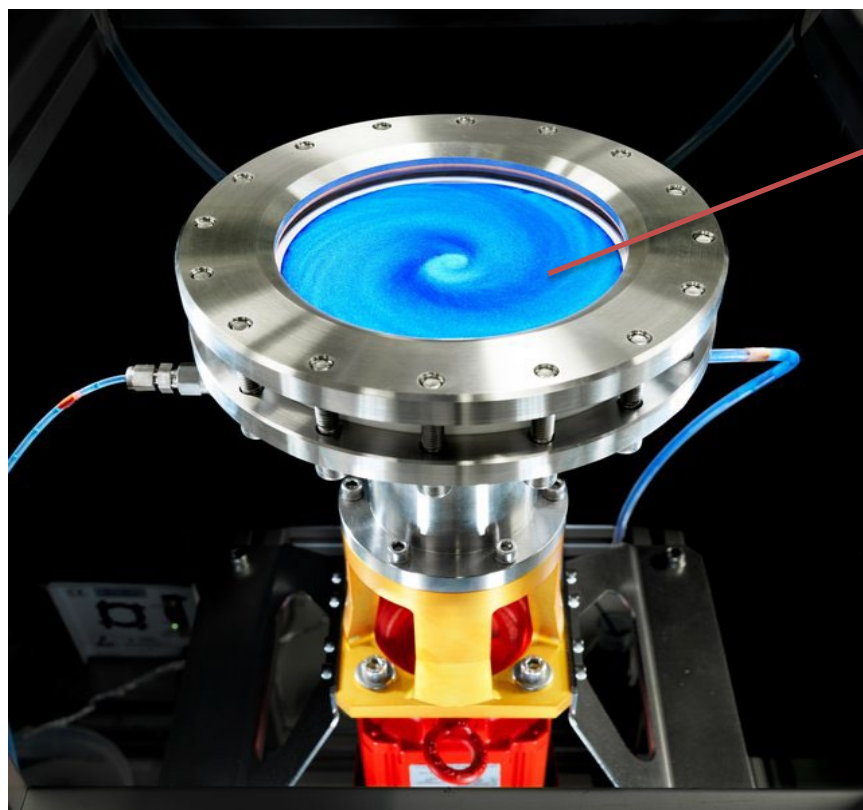
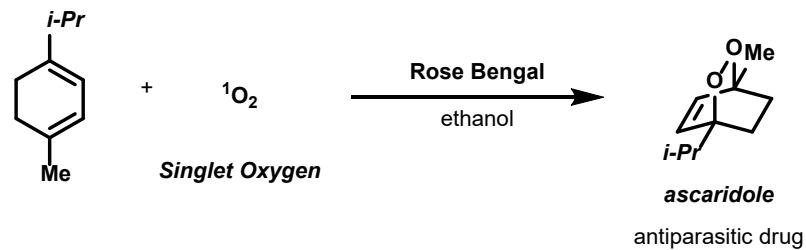
Scaling multiphase reaction conditions:



- Reactor = two stators and one rotating disc.
- Distance between rotor and stators is about 1-2 mm
- Rotational speed (1000-3000 rpm) induces high shear and thus provides high mass and heat transfer.

For a review: (i) F. Visscher, J. van der Schaaf, T. A. Nijhuis, J. C. Schouten, *Chem. Eng. Res. Des.* **2013**, *91*, 1923-1940. (ii) J. van der Schaaf, J. C. Schouten, *Curr. Opin. Chem. Eng.* **2011**, *1*, 84-88,

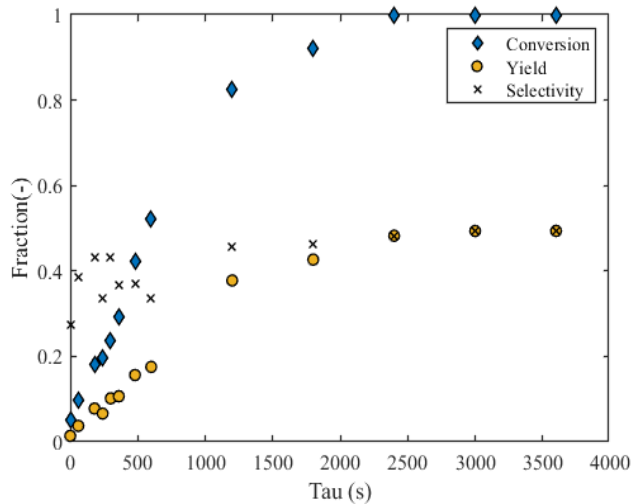
Rotor-Stator Spinning Disk Technology



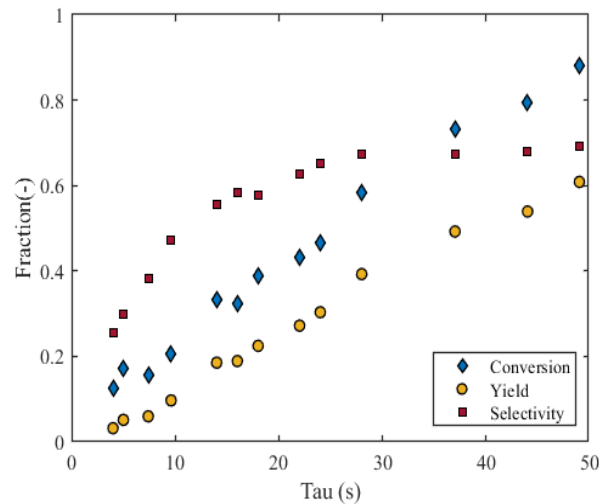
Quartz Window (Stator)

Rotor-Stator Spinning Disk Technology

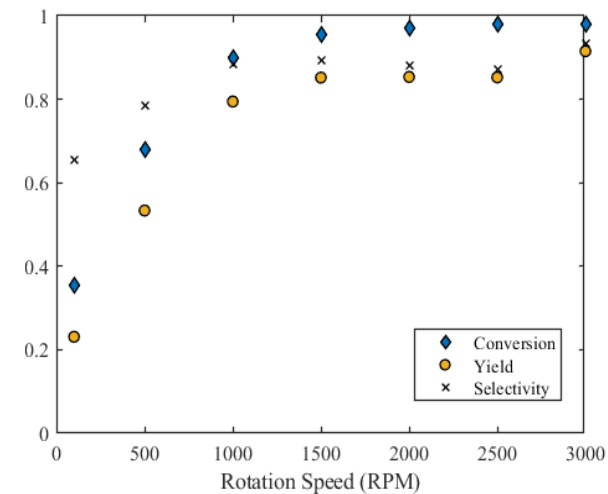
Batch ($V_R = 5$ mL)



Microflow
($id = 0.76$ mm, $V_R = 1$ mL)

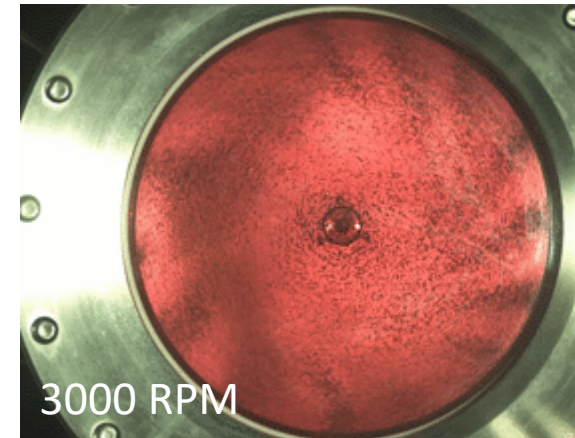
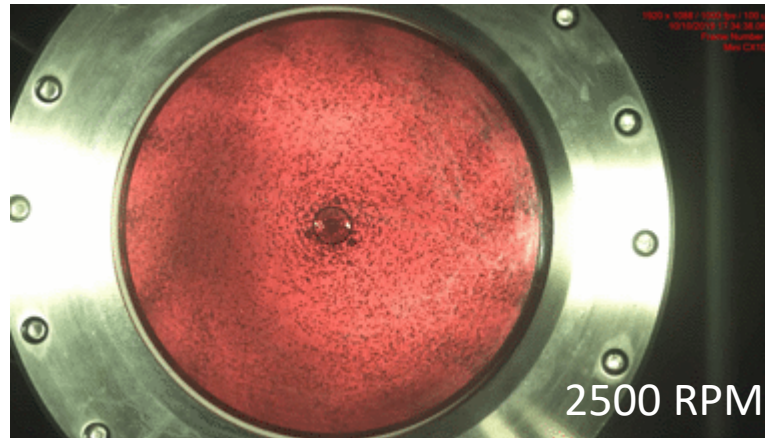
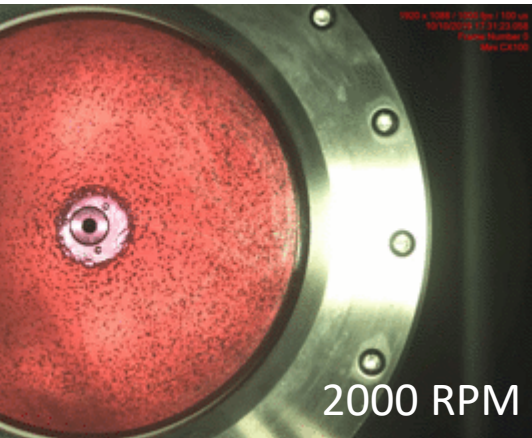
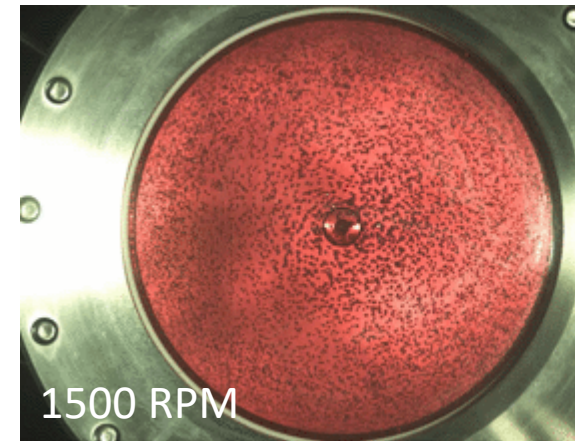
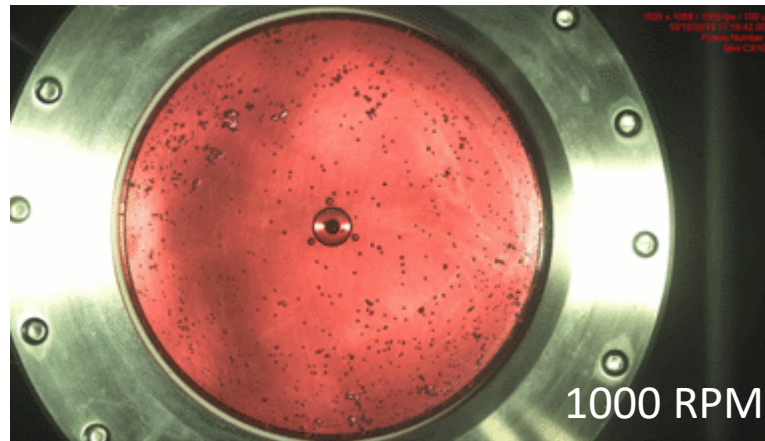
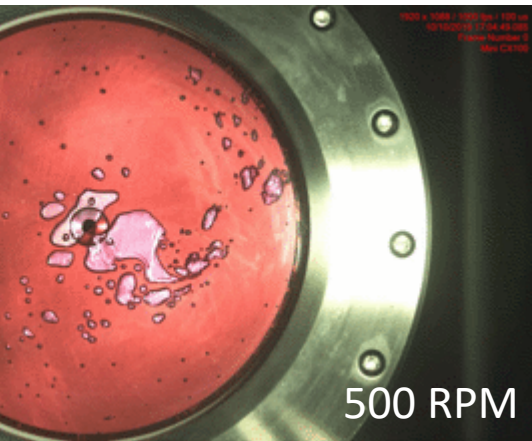


pSDR at 50 mL/min
($V_R = 28$ mL, $h = 2$ mm)



1. Selectivity goes up from 50% (batch) to 70% (flow) to 90-95% (pSDR).
2. Residence time is reduced from 40 minutes (batch) to 1 min (flow) to 27 seconds (pSDR)
3. Throughput increases from 0.375 mmol/h (batch) to 4.2 mmol/h (flow) to 270 mmol/h (pSDR)

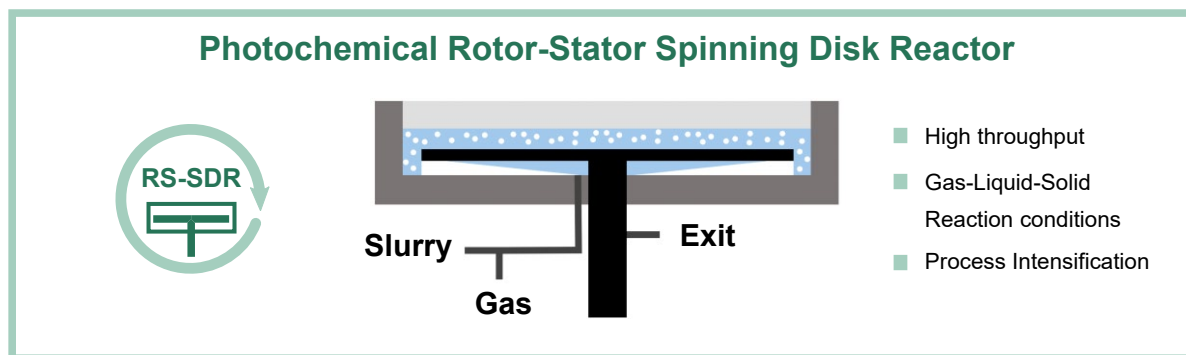
Rotor-Stator Spinning Disk Technology



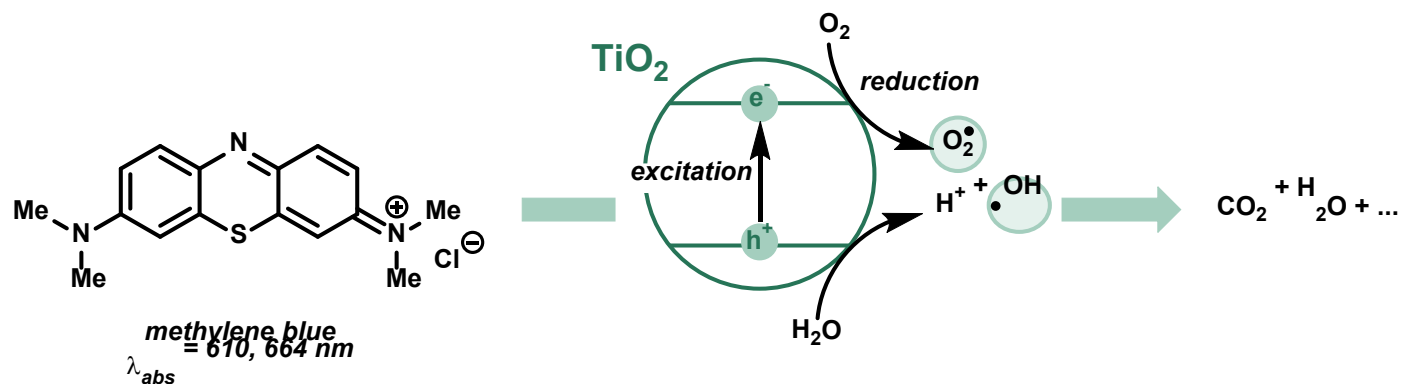
Photochemical Spinning Disk Reactor

Handling heterogeneous Photocatalysts

A Photochemical Rotor-Stator Spinning Disk Reactor enables scale up of complex heterogeneous photocatalytic reaction conditions



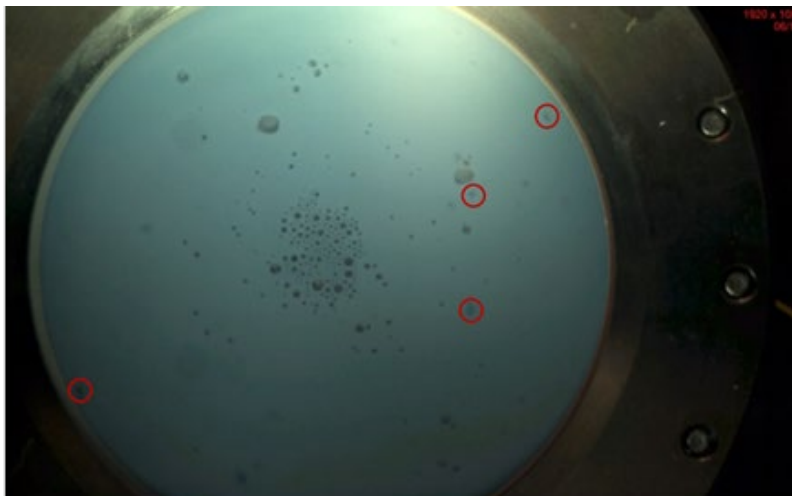
B Degradation of methylene blue enabled by titanium dioxide semiconductor photocatalysis



Photochemical Spinning Disk Reactor

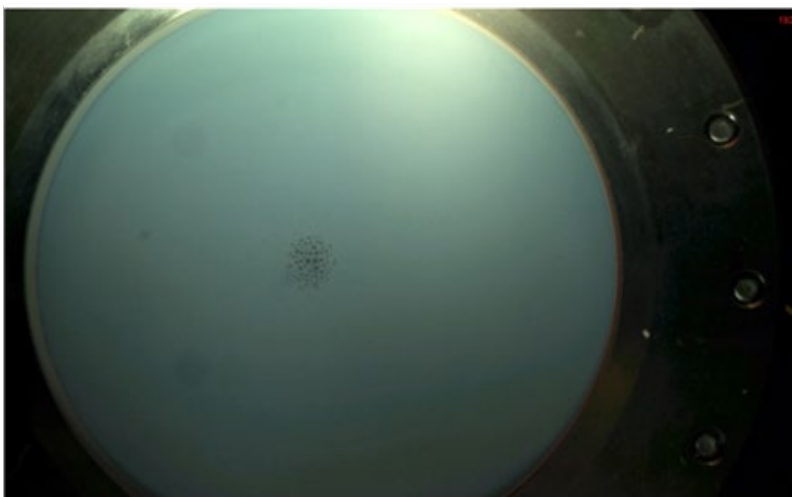
Handling heterogeneous Photocatalysts

A



500 RPM (red dots show particle agglomeration)

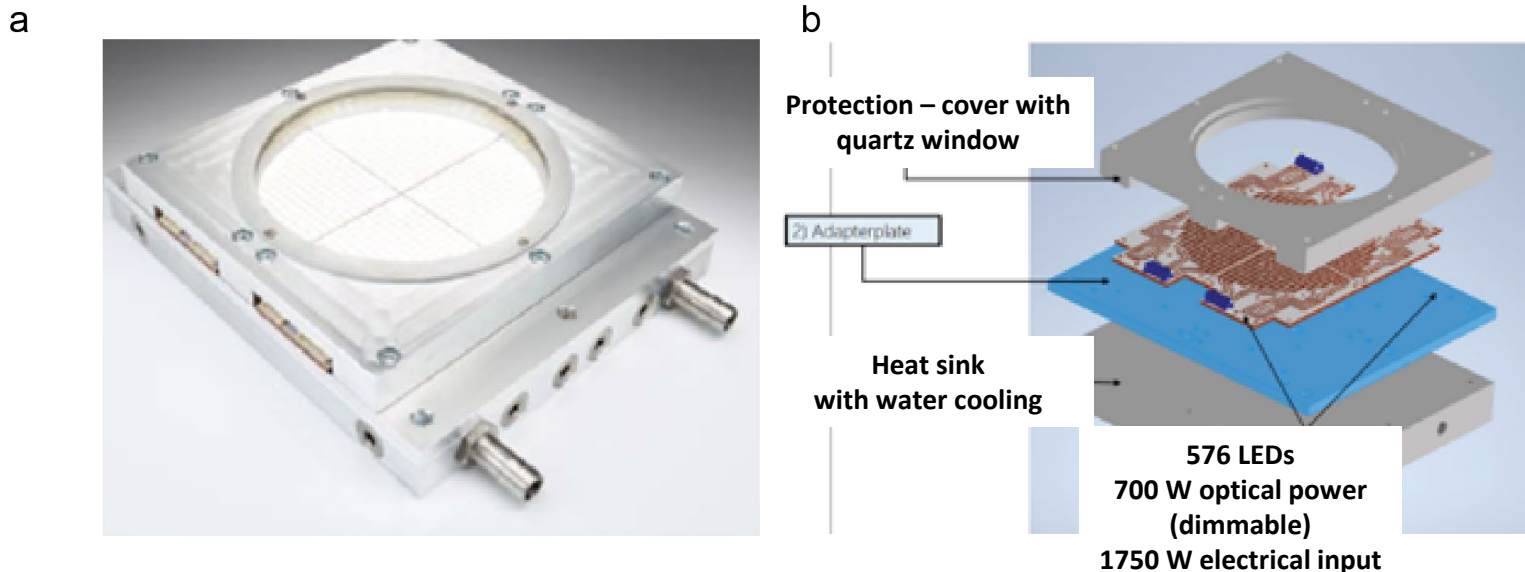
B



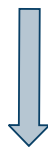
2000 RPM (10 mg/mL TiO₂)

Photochemical Spinning Disk Reactor

Matching with high power LEDs



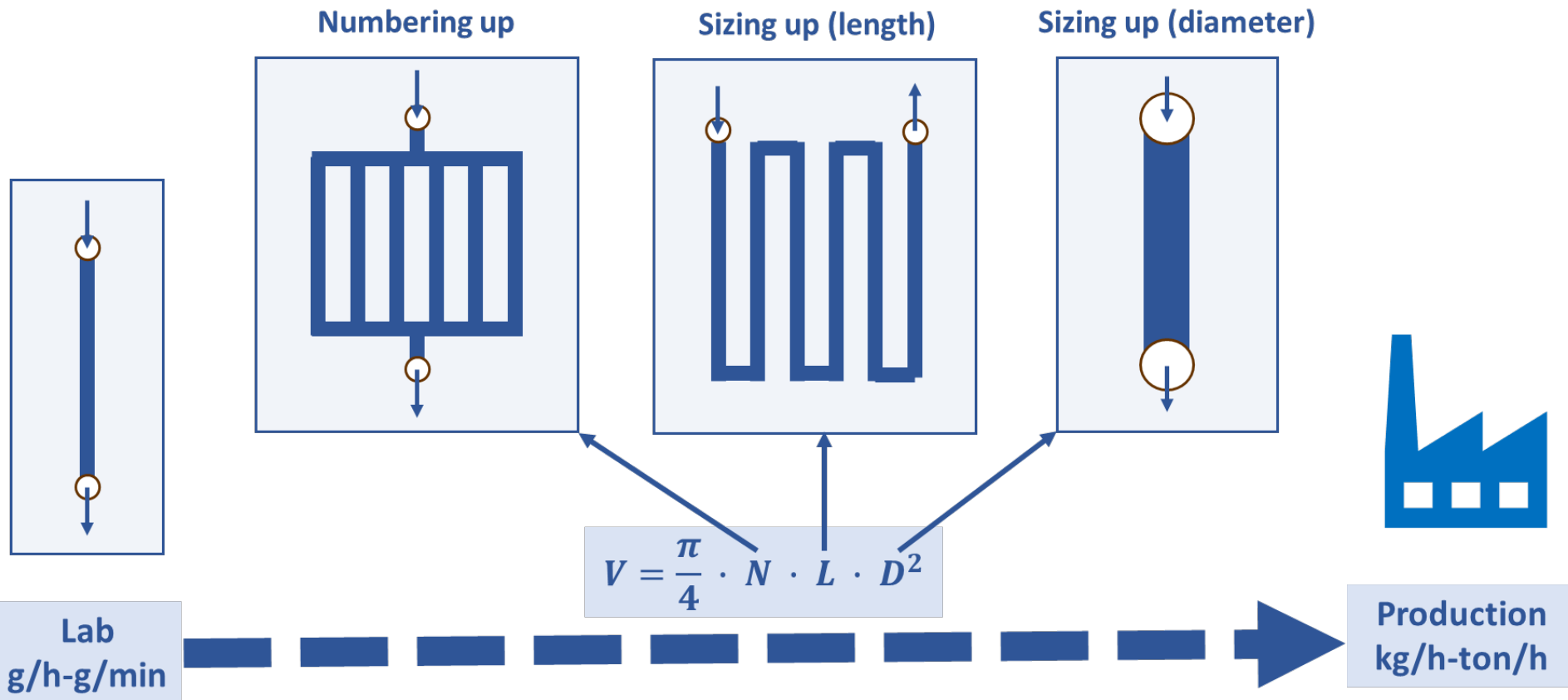
high light intensity enables to exploit excellent mass-transfer capacity of SDR



18 kg/day productivity for α -terpinene photooxidation

Photoredox Catalysis

scaling via flow technology



For reviews on scaling flow reactors: (i) Dong, Wen, Zhao, Kuhn, Noël, *Chem. Eng. Sci. X* **2021**, 100097. (ii) Donnelly, Baumann, *J. Flow Chem.* **2021**,11, 223-241. (iii) Berton, de Souza, Abdiaj, McQuade, Snead, *J. Flow Chem.* **2020**, 10, 73-92.

Synthetic methodology enabled by flow

Expanding Chemical Space

Photocatalytic sp^3 C–H functionalization via HAT



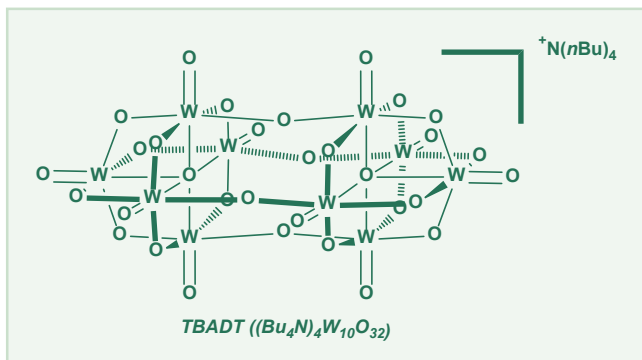
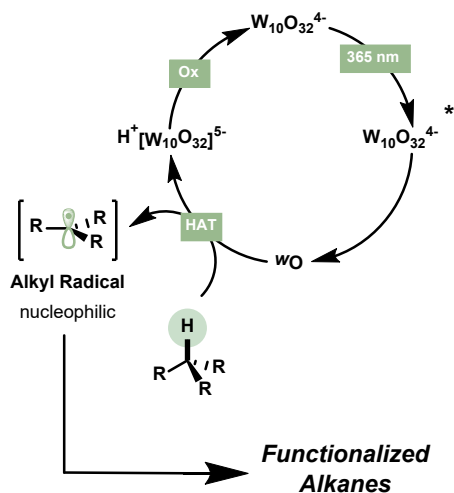
- generally unreactive bond
- activated and non-activated C–H bonds
- untapped synthetic potential

Photocatalytic sp^3 C–H functionalization via HAT

R–H

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- activated and non-activated C–H bonds
- untapped synthetic potential

Generation of radical intermediates
via Hydrogen Atom Transfer

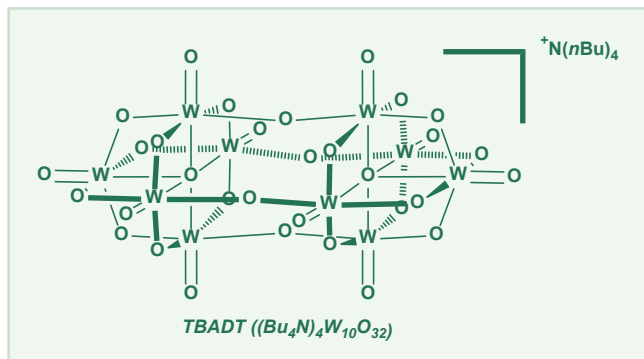
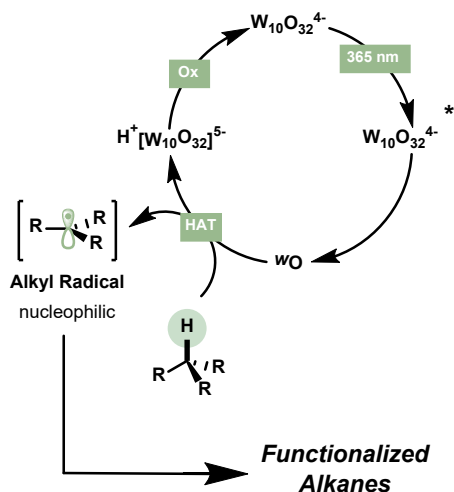


Photocatalytic sp^3 C–H functionalization via HAT



- generally unreactive bond
- activated and non-activated C–H bonds
- untapped synthetic potential

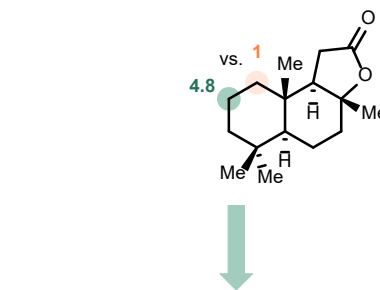
Generation of radical intermediates via Hydrogen Atom Transfer



Selectivity of decatungstate HAT photocatalysis

increased reactivity at:

- * allylic and benzylic positions
- * α positions compared to amines and ethers: $R-X-R$ with $X = O, N$
- * electron-rich heterocycles activate the benzylic position, vs. electron-poor heterocycles deactivate the benzylic position.
- * Inherent radical stability: tertiary > secondary > primary vs. steric hindrance due to large size of decatungstate



Allows for systematic structural tuning of biologically active molecules

This is an essential aspect of pharmaceutical research.

Photocatalytic HAT using gasses in flow



For a review: (i) Laporte, Masson, Zondag, Noël, *Angew. Chem. Int. Ed.* **2024**, *63*, e20231610.

(ii) Mallia, Baxendale, *Org. Process Res. Dev.* **2016**, *20*, 327–360.

Photocatalytic HAT using gasses in flow

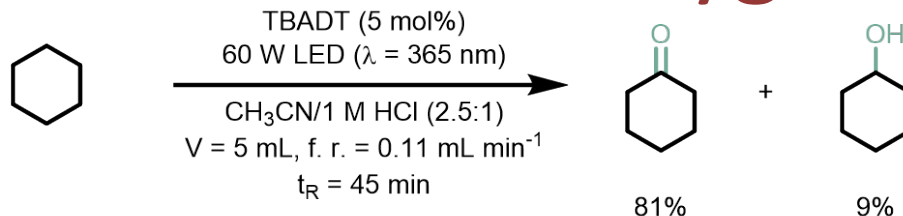


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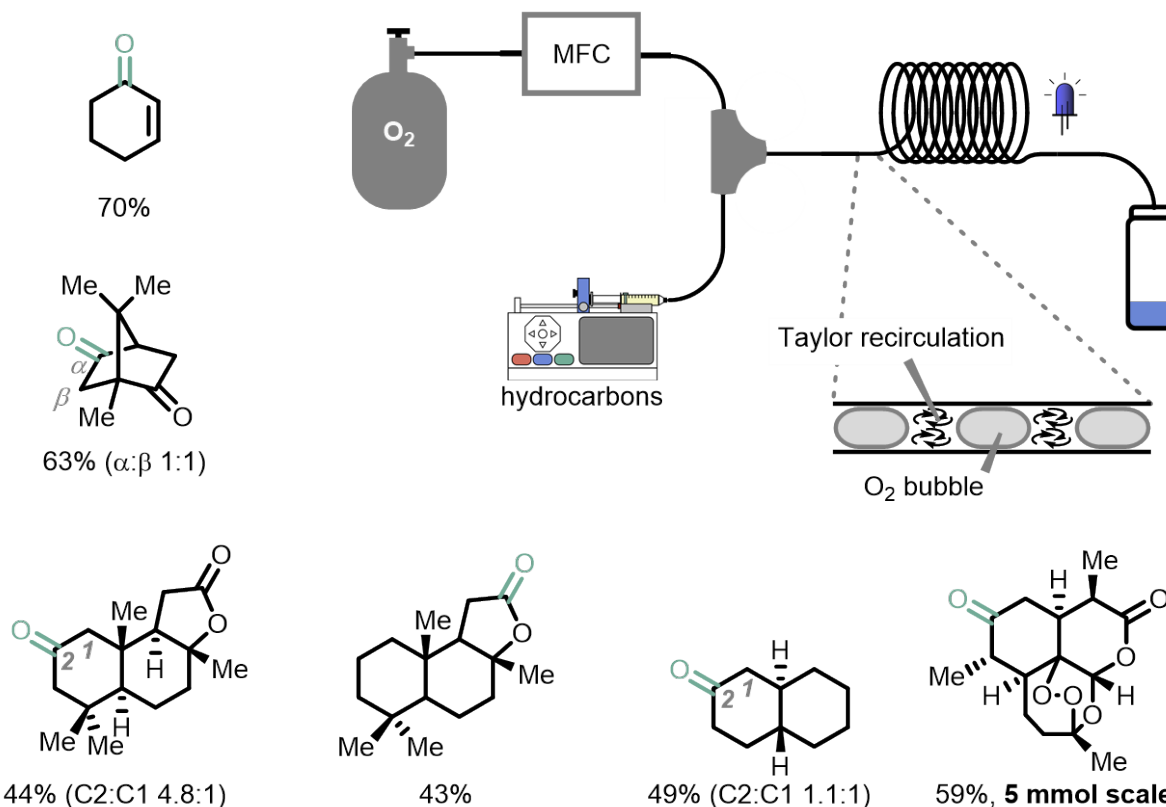
(ii) Mallia, Baxendale, *Org. Process Res. Dev.* **2016**, *20*, 327–360.

Photocatalytic sp^3 C–H oxidation

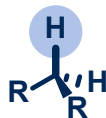
safe use of oxygen



Selected Examples



Most challenging problem in C–H functionalization

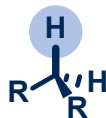


Gaseous Alkane

“Remaining unsolved, but increasingly important due to the production of shale gas, is the original goal: the mild and selective conversion of methane and light hydrocarbons to functionalized feedstocks”

John F. Hartwig, *JACS* **2016**, 138, 2-24.

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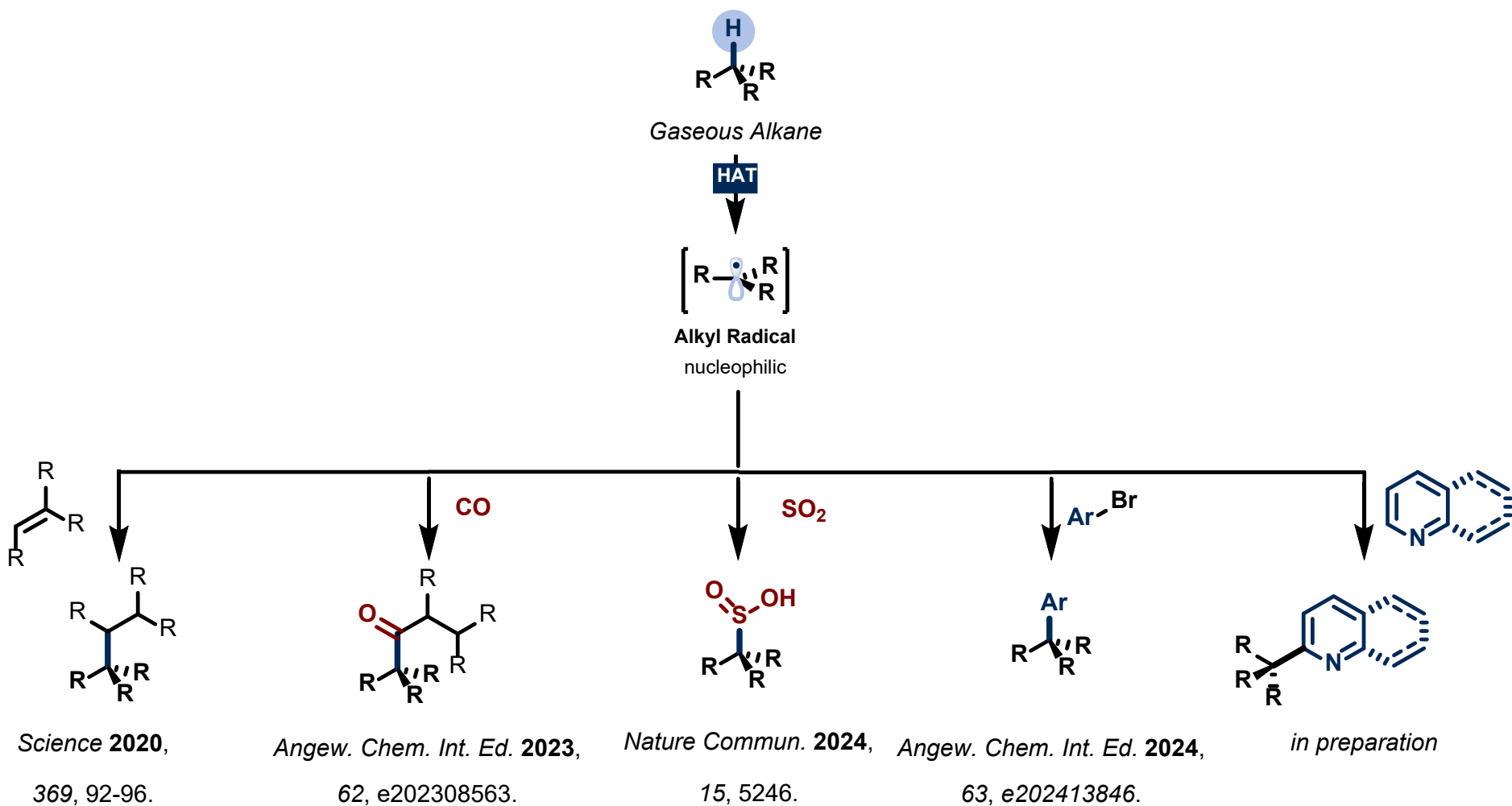
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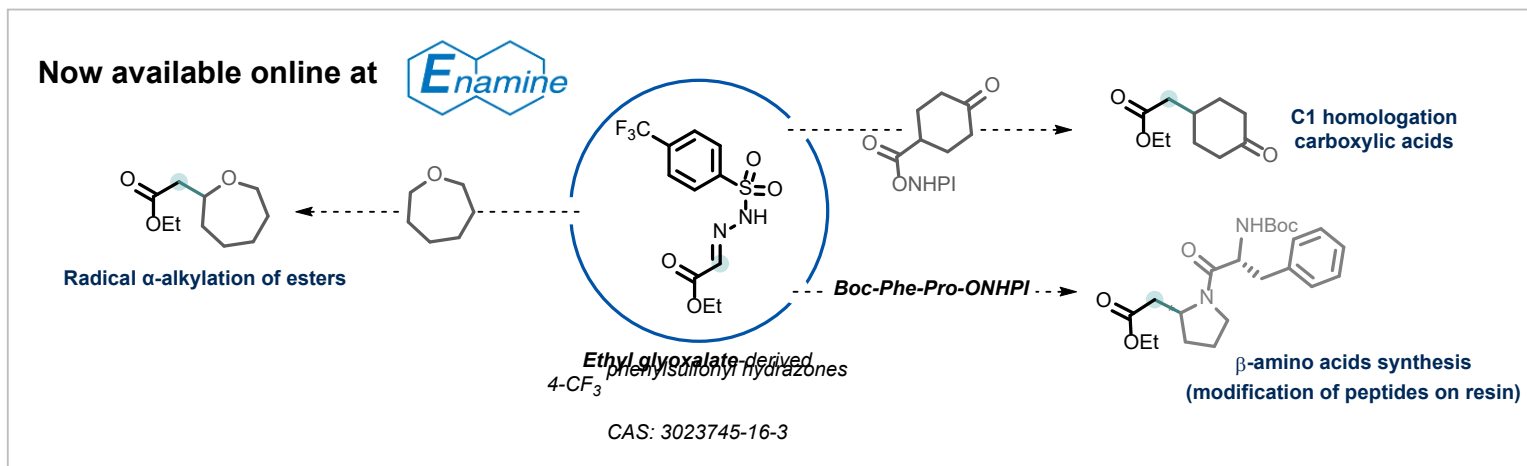
“[Natural gas] is a vast, low-cost feedstock of hydrocarbons that remains untapped as a raw material, simply because there has been no easy way to turning it into synthetically useful compounds”

Robert G. Bergman, *Nature* **2007**, 138, 391-392.

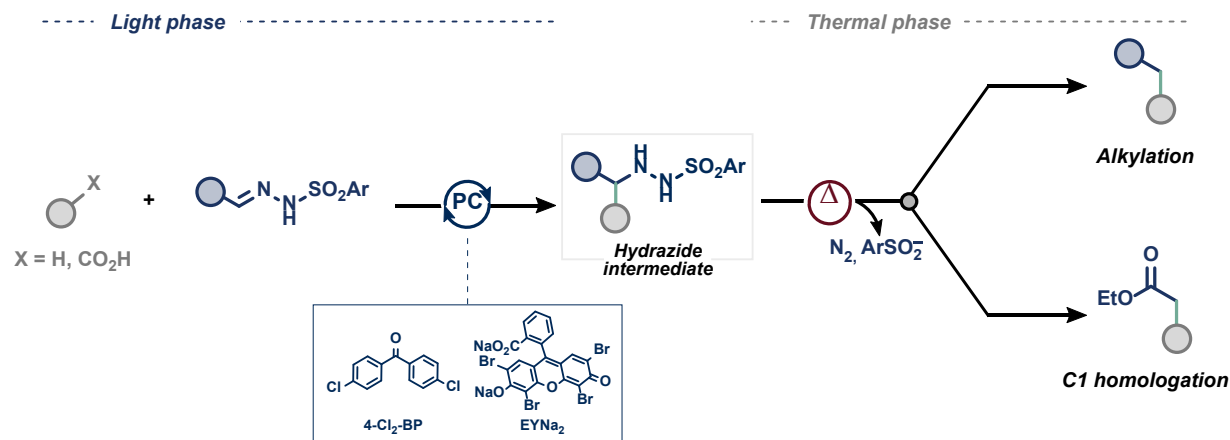
Photocatalytic sp^3 C–H functionalization of light alkanes



Sulfonyl hydrazone, a versatile building block



Via a one-pot, two-step process:



For HAT: Pulcinella, Bonciolini, Lukas, Sorato, Noël, *Angew. Chemie Int. Ed.* **2023**, 62, e202215374.

For decarboxylative coupling: Bonciolini, Pulcinella, Leone, Schirotti, Luguera Ruiz, Sorato, Dubois, Gopalakrishnan, Masson, Della Ca', Protti,

Fagnoni, Zysman-Colman, Johansson, Noël, *Nature Commun.*, **2024**, 15, 1509.

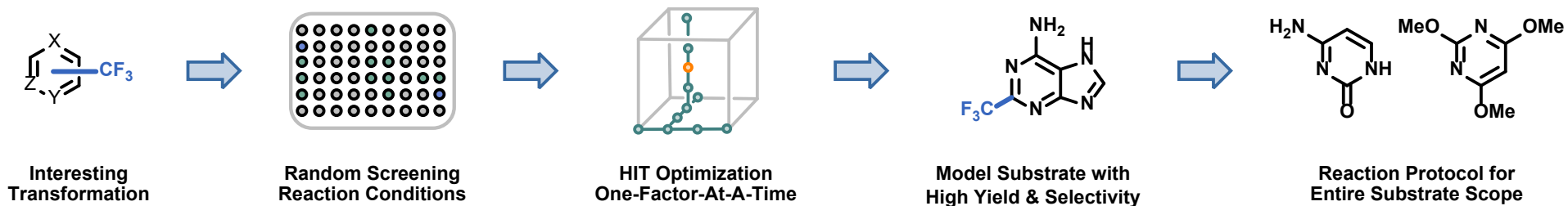
Reaction Optimization

Streamlining and optimizing a compound trace into a widely applicable synthetic method is both demanding and time-intensive.

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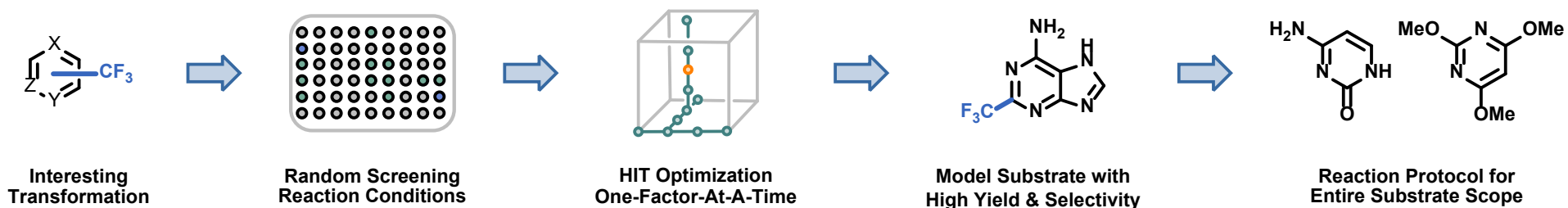
Typical strategy for synthetic methodology development:



Reaction Optimization

Streamlining and optimizing a compound trace into a widely applicable synthetic method is both demanding and time-intensive.

Typical strategy for synthetic methodology development:



Most substrates have suboptimal yields.

Reaction Optimization

If most substrates in a scope have suboptimal yields.



Why not let a machine do the work?



Photocatalysis offers distinct organic synthesis methods but faces significant challenges.



Technological complexity

- setup variability
- photon absorption
- limited scalability



Challenges

- multiparametric optimization
- poor reproducibility
- challenging scale up



Chemical complexity

- limited mechanistic understanding
- complex photophysics
- one-factor-at-a-time optimization

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Challenges

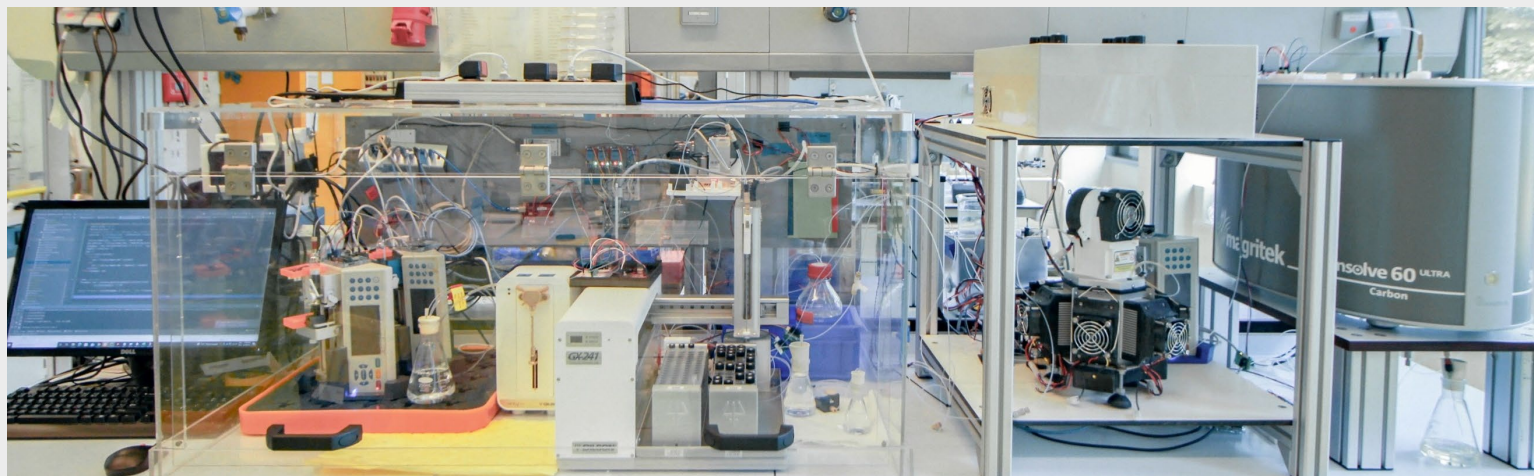
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RoboChem: multipurpose robotic platform for the self-optimization, intensification and scale-up of photocatalytic transformations



Bayesian Optimization
& Graphical User Interface

Sample Preparation

Photochemical Reactor

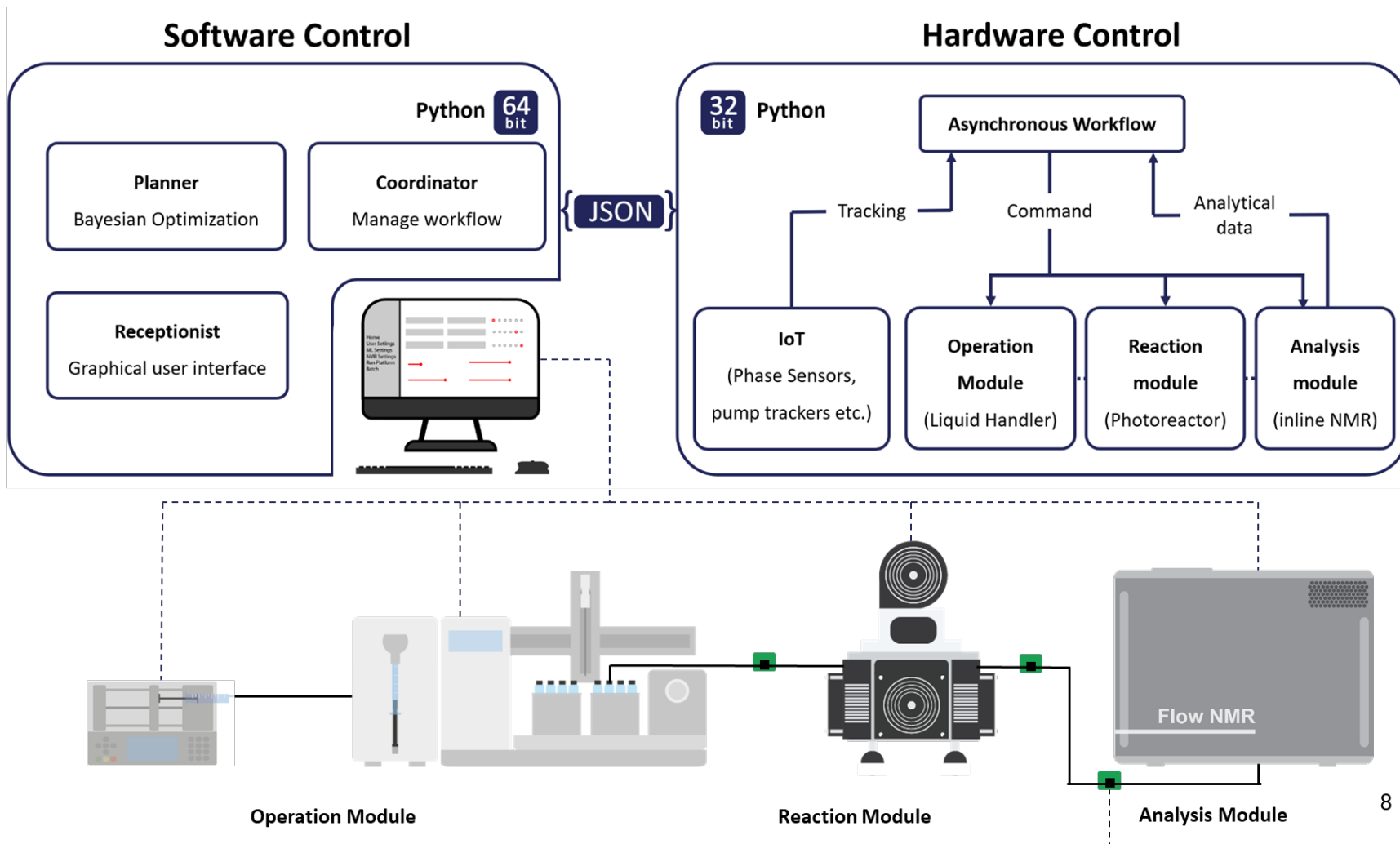
Inline NMR

Closed-loop, multiobjective optimization

- fast optimization
- reproducible
- captures complex intercorrelations
- generation of detailed datasets
- scalable process

RoboChem

An all-in-one robotic platform



RoboChem

An all-in-one robotic platform

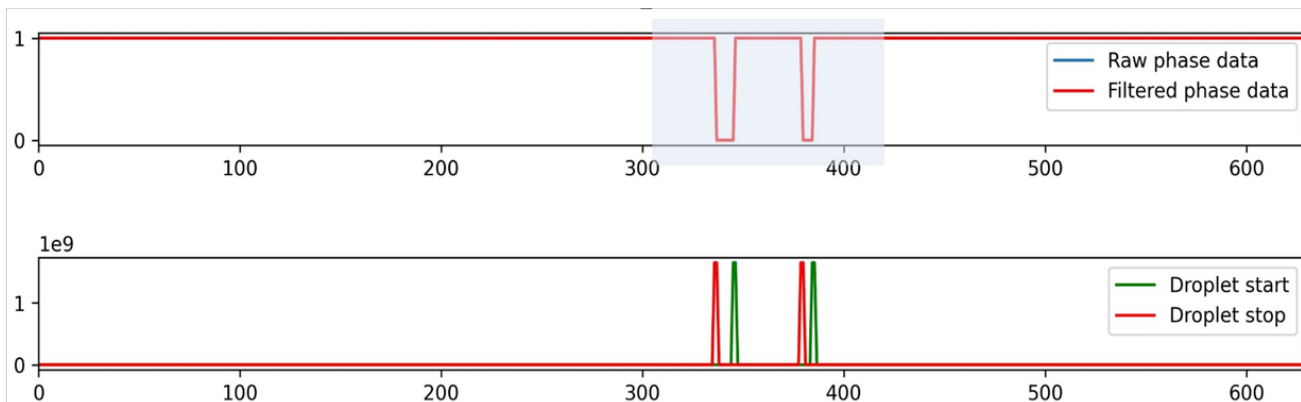
Carrier solvent

N₂

Reaction droplet

N₂

Carrier solvent

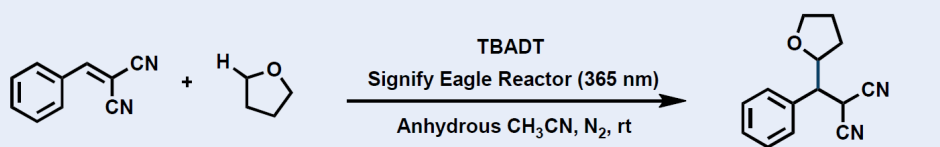


Phase Sensor

RoboChem

An all-in-one robotic platform

C–H alkylation via photocatalytic HAT

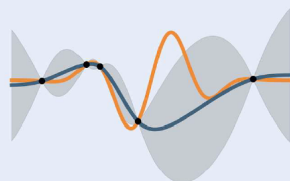


3.7 mmol scale: 99% yield

Optimization Variables (5)

- Substrate Concentration (0.05-0.2M)
- THF loading (1-18 equiv)
- TBADT loading (0.5-3 mol%)
- Residence time (120-360 s)
- Light Intensity (5 - 144 W)

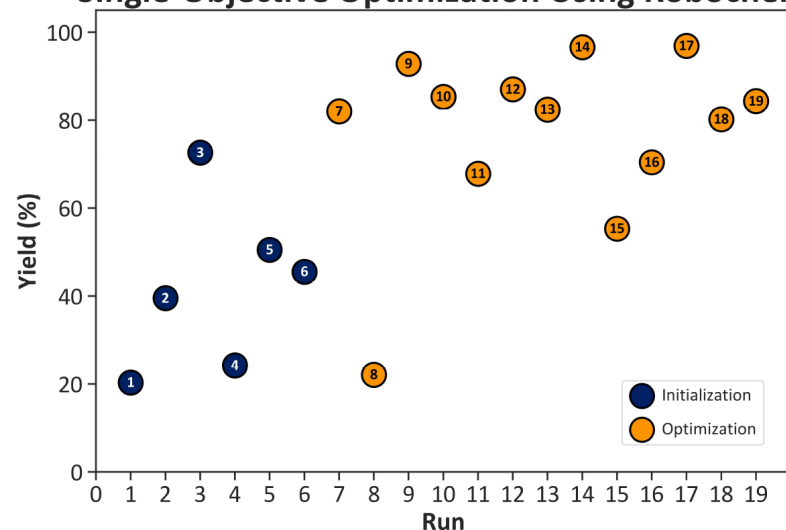
Bayesian Optimization Algorithm



Objective Functions (1)

↑ Yield (%)

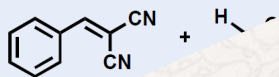
Single-Objective Optimization Using RoboChem



RoboChem

An all-in-one robotic platform

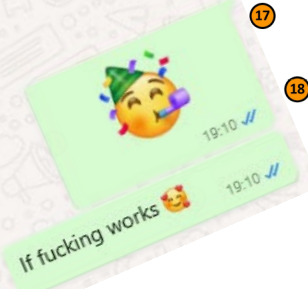
C–H alkylation via photoredox



How about them yields? 😊

Subs
7
TBAI
Resid
Light Ir.

Robochem



● Initialization
● Optimization

0 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19
Run

Trifluoromethylthiolation via HAT

C–H trifluoromethylthiolation via photocatalytic Hydrogen Atom Transfer



Optimization Variables (5)

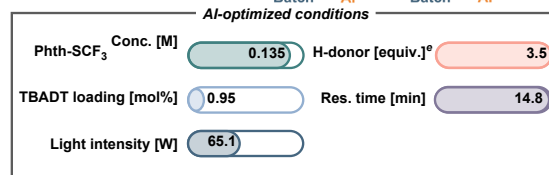
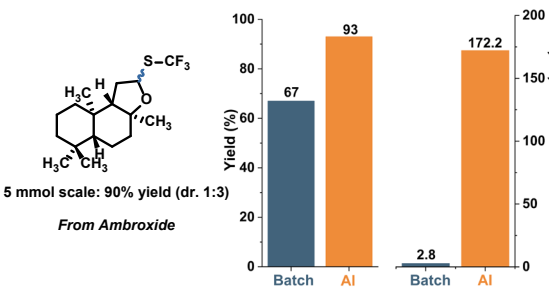
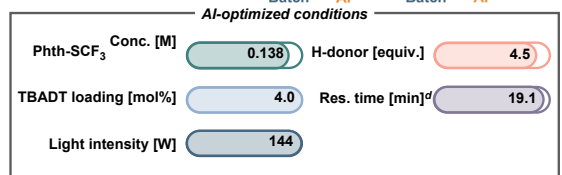
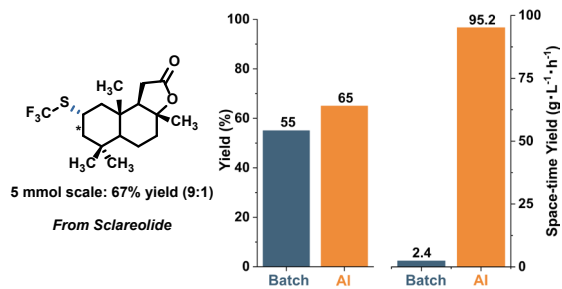
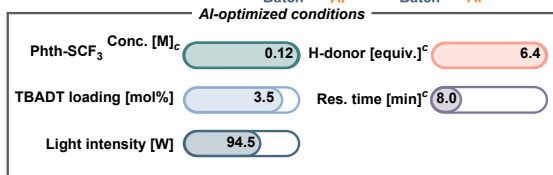
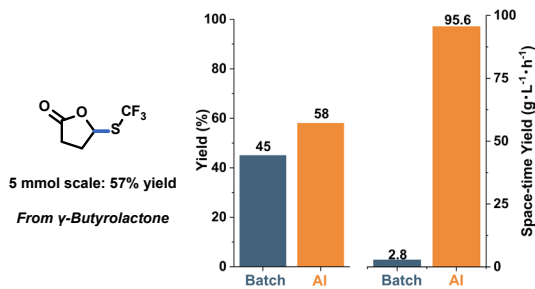
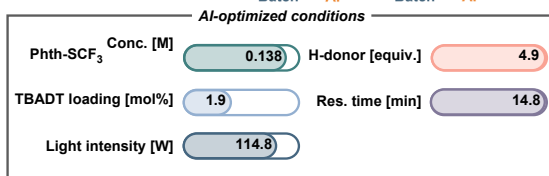
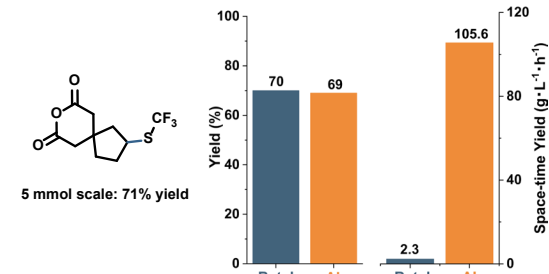
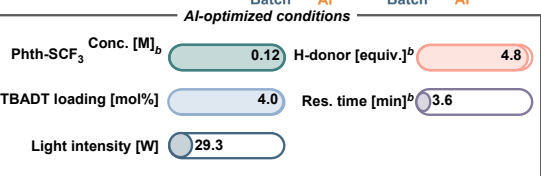
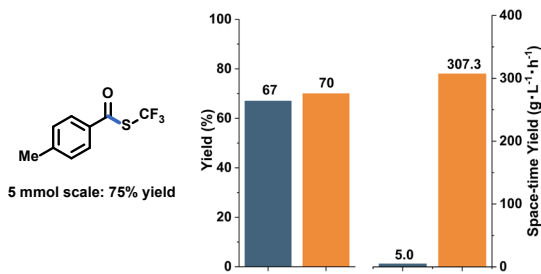
Phth-SCF₃ Conc. (0.05-0.15 M)
 H-donor loading (1-5 equiv)
 TBADT loading (0.5-4 mol%)
 Res. time (2-15 min)
 Light intensity (0-144 W)

Bayesian Optimization



Objective Functions (2)

↑ Yield (%)
 ↑ Space-time yield (g·L⁻¹·h⁻¹)



Original work: Schirmer, Rolka, Karl, Holzhausen, König, *Org. Lett.* **2021**, *23*, 5729–5733.

Slattery, Wen, Tenblad, Pintossi, Orduna, den Hartog, Noel, *Science* **2024**, *383*, eadj1817.

Oxytrifluoromethylation via SET

Oxytrifluoromethylation via photocatalytic Single Electron Transfer



Optimization Variables (5)

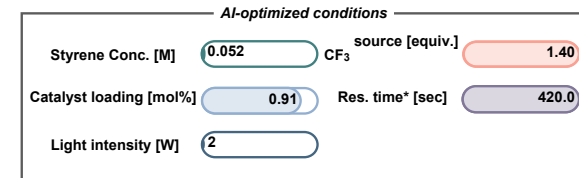
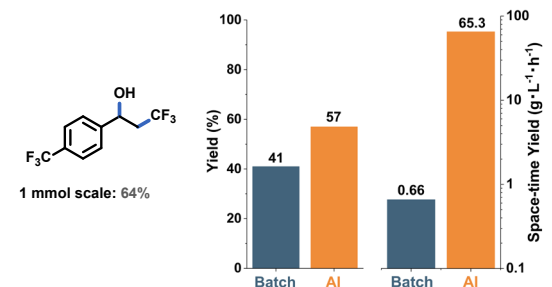
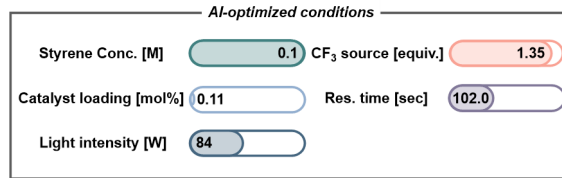
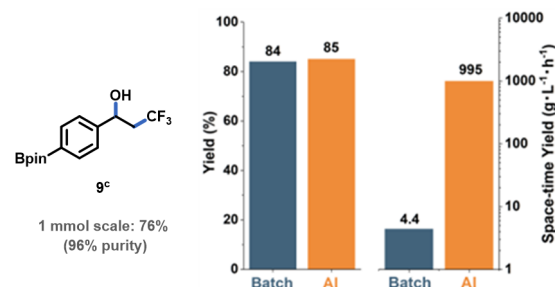
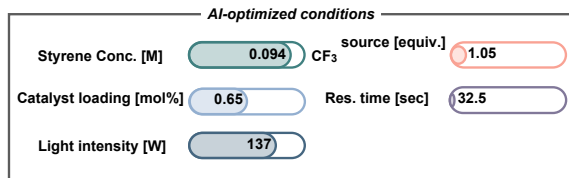
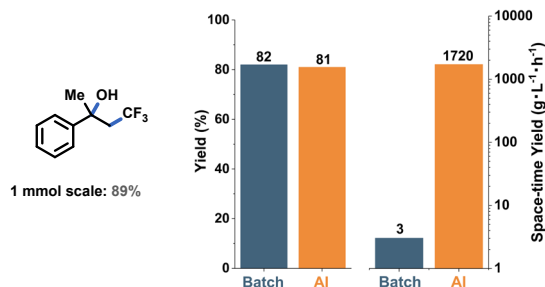
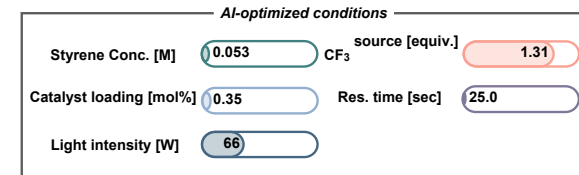
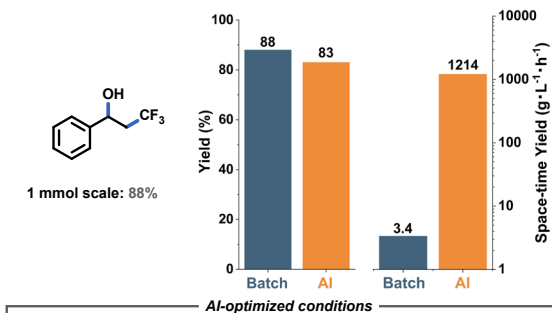
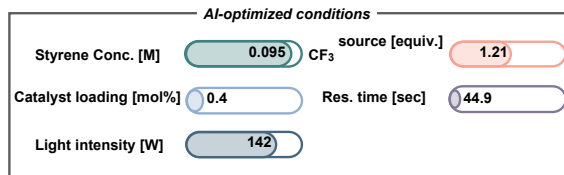
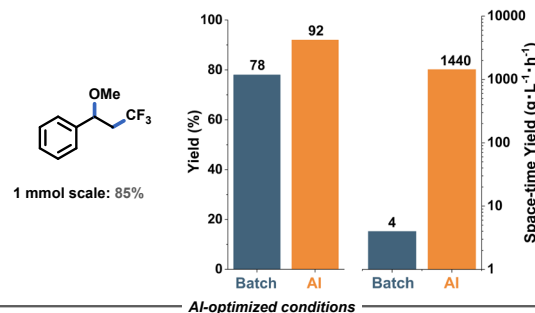
Bayesian Optimization

Objective Functions (2)

Styrene Conc. (0.05-0.10 M)
 CF₃ source loading (1-1.4 equiv)
 catalyst loading (0.1-1 mol%)
 Res. time (20-300 sec)
 Light intensity (0-182 W)



↑ Yield (%)
 ↑ Space-time yield (g·L⁻¹·h⁻¹)

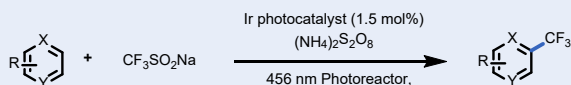


Original work: Yasu, Koike, Akita, *Angew. Chem. Int. Ed.* **2012**, *51*, 9567-9571.

Slattery, Wen, Tenblad, Pintossi, Orduna, den Hartog, Noel, *Science* **2024**, *383*, eadj1817.

Trifluoromethylation via SET

Aryl trifluoromethylation via photocatalytic Single Electron Transfer

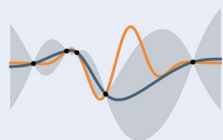


Optimization Variables (5)

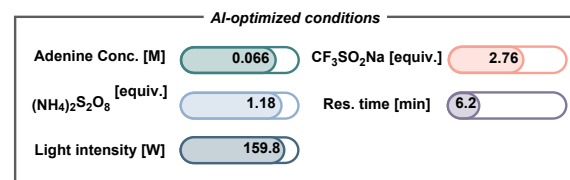
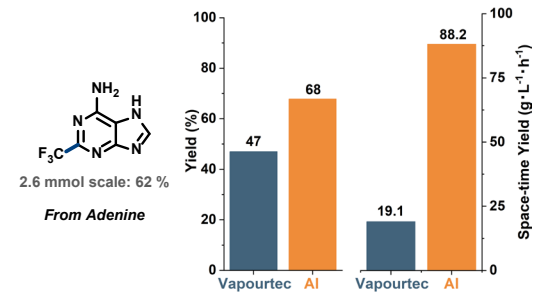
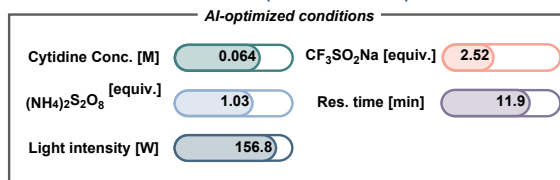
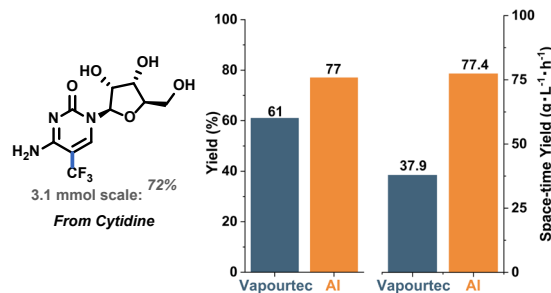
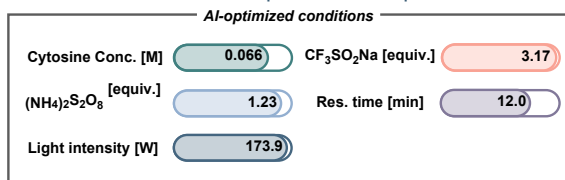
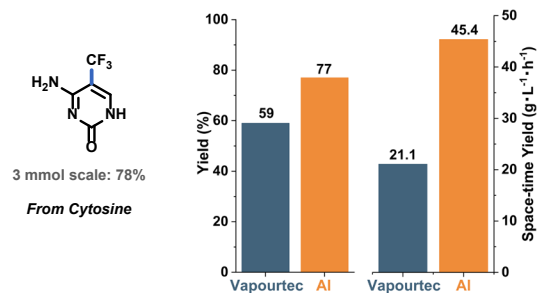
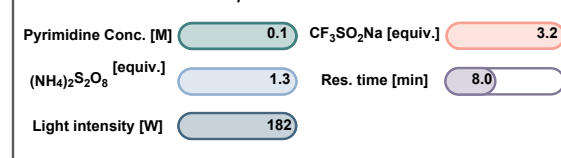
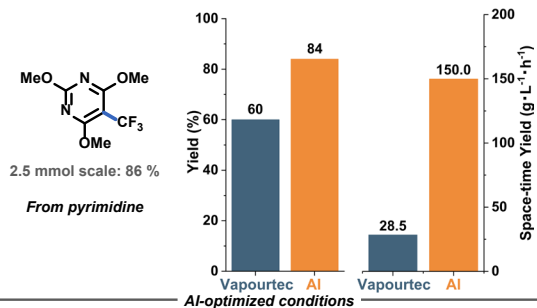
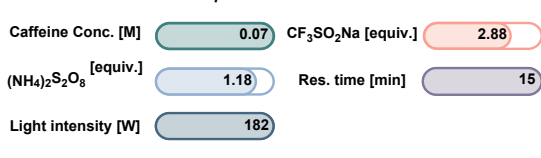
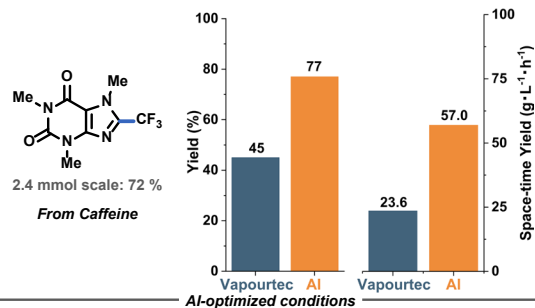
Bayesian Optimization

Objective Functions (2)

Heteroarene Conc. (0.05-0.07 M)
 $\text{CF}_3\text{SO}_2\text{Na}$ loading (2-3.2 equiv)
 $(\text{NH}_4)_2\text{S}_2\text{O}_8$ loading (0.5-1.3 equiv)
 Residence time (5-15 min)
 Light intensity (0-182 W)



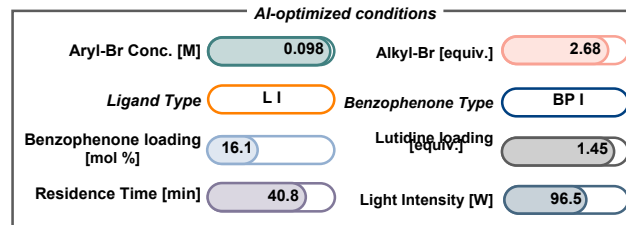
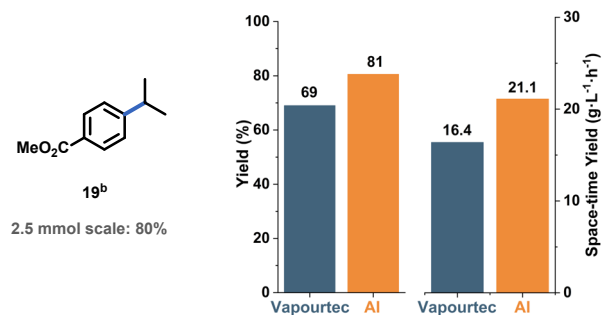
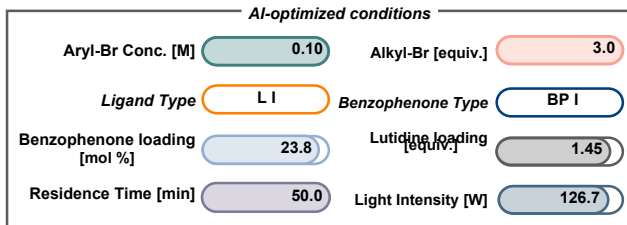
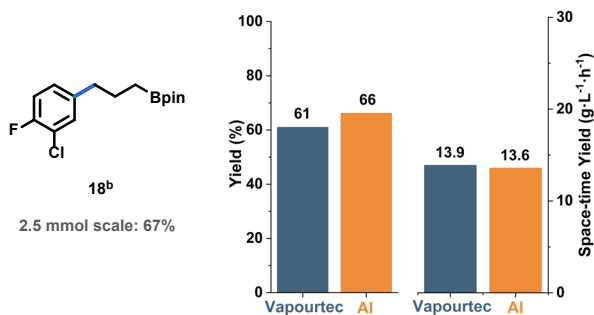
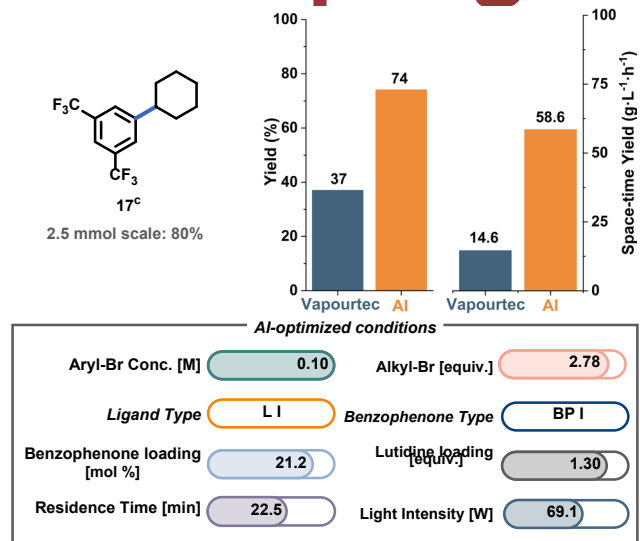
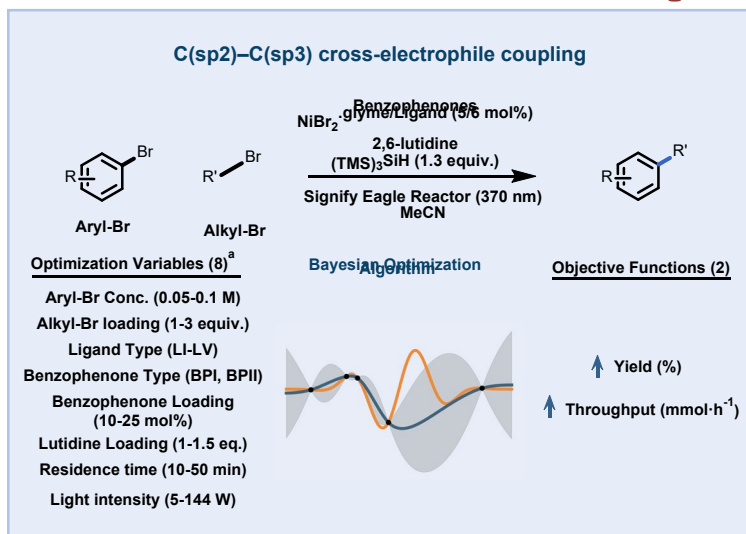
↑ Yield (%)
 Space-time yield ($\text{g}\cdot\text{L}^{-1}\cdot\text{h}^{-1}$)



Original work: Abdiaj, Bottecchia, Alcazar, Noël, *Synthesis* **2017**, 49, 4978-4985.

Slattery, Wen, Tenblad, Pintossi, Orduna, den Hartog, Noel, *Science* **2024**, 383, eadj1817.

Cross-Electrophile Coupling



Original work: Luridiana, Mazzarella, Capaldo, Rincon, Garcia-Losada, Mateos, Frederick, Nuno, Buma, Noël, *ACS Catal.* **2022**, *12*, 11216–11225.

Slattery, Wen, Tenblad, Pintossi, Orduna, den Hartog, Noel, *Science* **2024**, *383*, eadj1817.

Is chemistry ready for an AI revolution?

Is chemistry ready for an AI revolution?

Machine Learning May Sometimes Simply Capture Literature Popularity Trends: A Case Study of Heterocyclic Suzuki–Miyaura Coupling

Wiktor Beker, Rafał Roszak, Agnieszka Wołos, Nicholas H. Angello, Vandana Rathore, Martin D. Burke,* and Bartosz A. Grzybowski*

 Cite This: *J. Am. Chem. Soc.* 2022, 144, 4819–4827

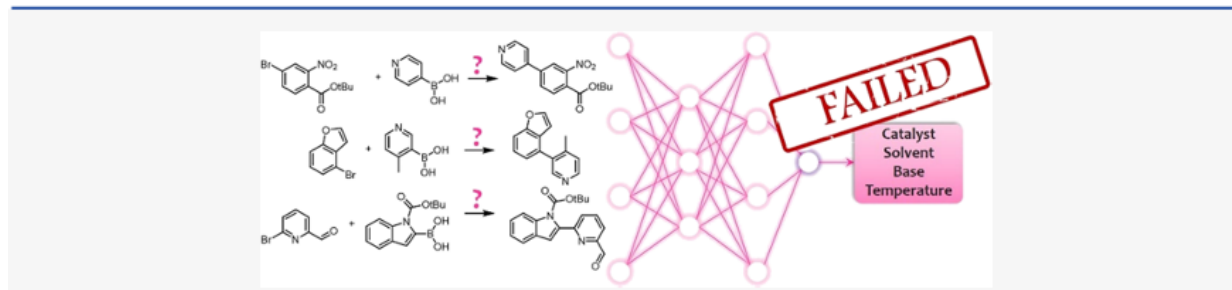
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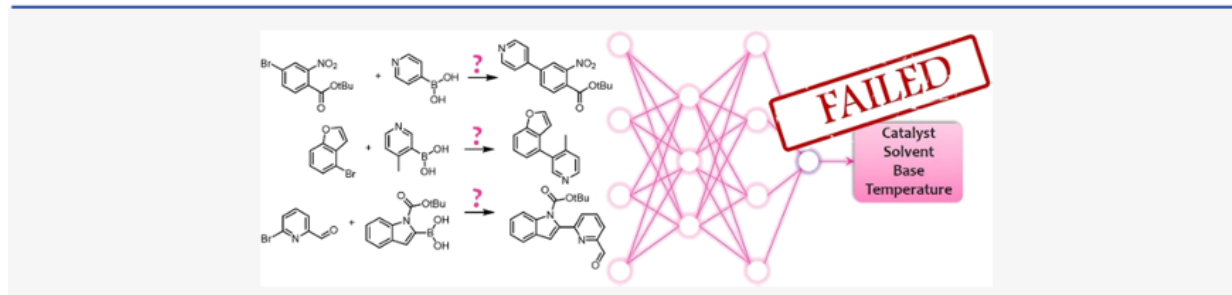
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Caused by

- Subjective preferences in selecting reaction conditions by chemists
- Lack of reliable and standardized data, including lack of negative data.

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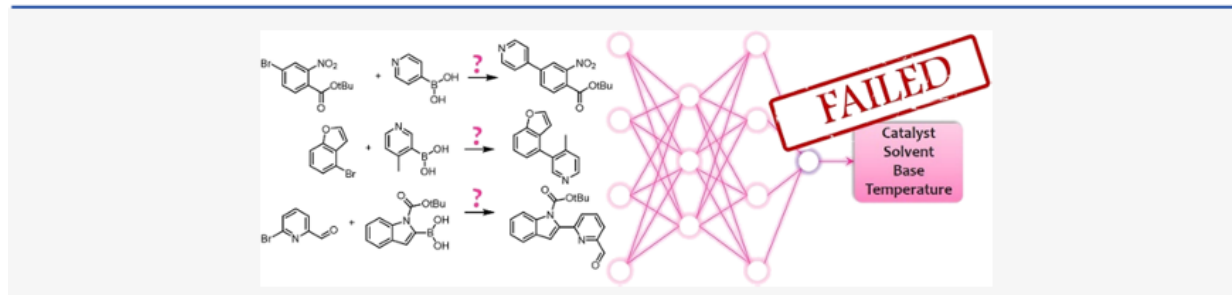
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Caused by

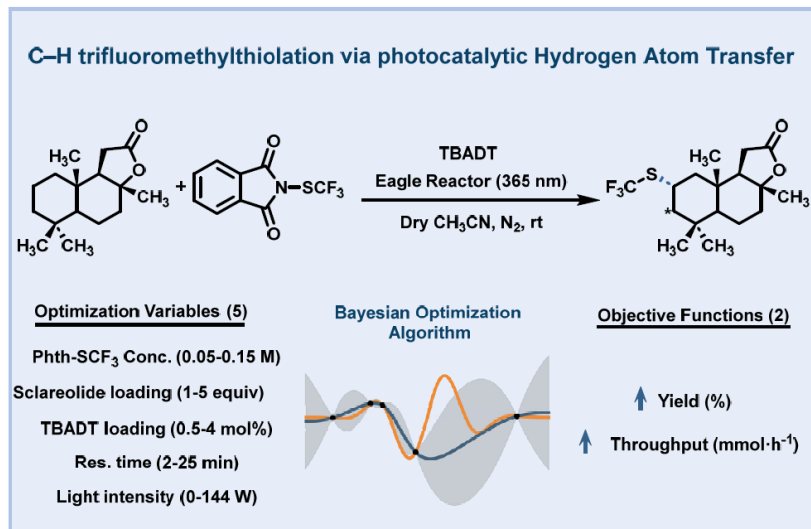
- Subjective preferences in selecting reaction conditions by chemists
- Lack of reliable and standardized data, including lack of negative data.



Fundamental flaw in the current synthetic literature (!)

Grzybowski et al., *J. Am. Chem. Soc.* 2022, 144, 4819–4827.

Digitization of Chemistry



Clean, detailed and reproducible datasets

- No mass, heat or photon transfer issues
- No human error
- Positive and negative data available

Table S15. Experimental conditions and results for trifluoromethylthiolation campaign of sclareolide.

run	Phth-SCF ₃ Conc. (M)	Sclareolide loading (equiv.)	TBADT loading (mol%)	Residence time (min)	Light intensity (W)	Yield (%)	Throughput (mmol/hr)
1	0.100	4.00	4.00	15.0	144	58.6	0.668
2	0.064	4.42	1.75	5.1	117	21.2	0.457
3	0.132	1.95	3.35	8.8	104	50.4	1.282
4	0.125	1.88	1.38	9.0	108	51.6	1.226
5	0.062	1.63	2.44	12.3	1	1.8	0.015
6	0.100	2.75	2.25	9.0	108	57.1	1.084
7	0.065	3.77	3.18	10.9	84	51.5	0.523
8	0.138	4.06	3.56	4.5	36	5.4	0.282
9	0.075	1.88	3.13	9.0	36	4.2	0.059
10	0.100	3.42	2.72	6.0	32	3.5	0.100
11	0.141	2.68	0.90	11.2	135	32.2	0.690
12	0.149	4.47	3.97	19.1	92	47.1	0.630
13	0.093	1.38	2.79	9.0	144	25.8	0.456
14	0.100	4.28	3.94	13.4	122	50.1	0.641
15	0.138	4.50	4.00	19.8	144	65.3	0.776
16	0.138	2.31	2.69	9.4	108	4.7	0.119
17	0.145	3.21	4.00	19.1	97	60.7	0.789
18	0.131	4.45	3.93	13.6	105	58.4	0.961
19	0.128	1.68	1.22	17.2	61	31.7	0.403
20	0.072	3.26	1.87	4.5	142	11.2	0.304
21	0.121	2.00	3.13	7.3	36	8.7	0.248
22	0.091	5.00	1.37	8.9	0	0.0	0.000
23	0.092	4.75	0.72	18.9	63	23.0	0.190
24	0.115	4.24	3.94	13.7	64	52.0	0.750
25	0.082	4.53	1.21	15.8	85	36.6	0.327
26	0.119	4.88	2.69	13.4	132	52.3	0.791

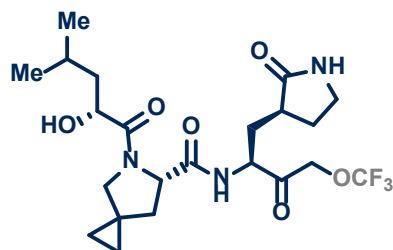
Slattery, Wen, Tenblad, Pintossi, Orduna, den Hartog, Noel, *Science* **2024**, *383*, eadj1817.

For a review on self-driving labs, see: Bailey, Slattery, Savino, Noel, *Matter* **2024**, *7*, 2382–2398.

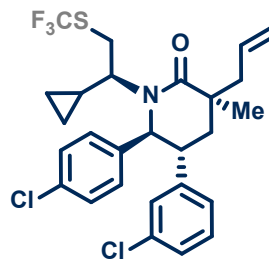
On-demand Generation Fluorinated Reagents

Emerging fluorinated moieties

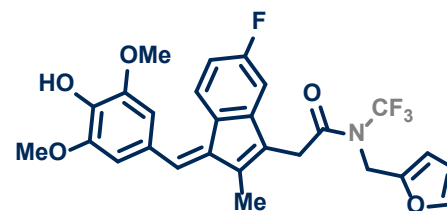
Growing interest into trifluoromethyl groups attached to heteroatoms



CMX990
SARS-CoV-2 protease inhibitor



WO2011153509
MDM2 inhibitor intermediate

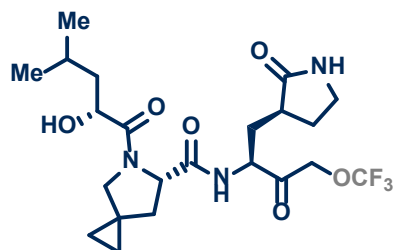


WO2016100542
RAS Gtpase inhibitor

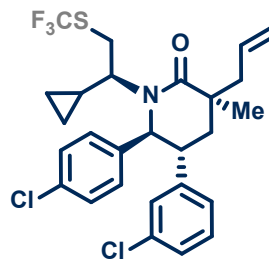
- New chemical space by the XCF_3 motifs
- Fine tuning of the molecular properties
- XCF_3 are gaining momentum as an alternative to $C-CF_3$ fragments

Emerging fluorinated moieties

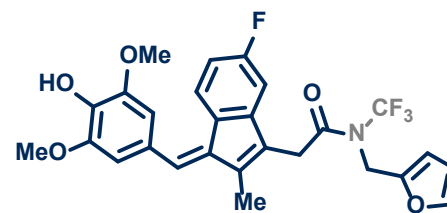
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CMX990
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WO2011153509
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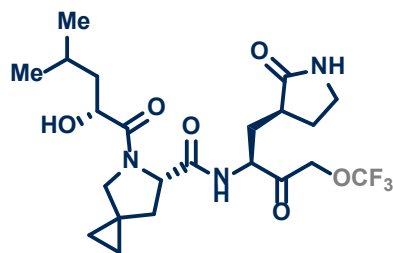
WO2016100542
RAS Gtpase inhibitor

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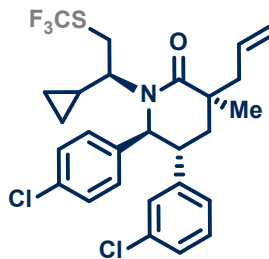
But how to make them?

Emerging fluorinated moieties

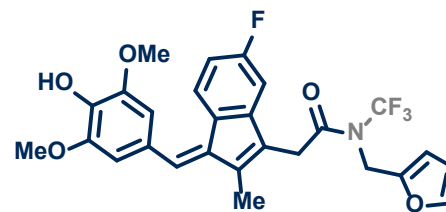
Growing interest into trifluoromethyl groups attached to heteroatoms



CMX990
SARS-CoV-2 protease inhibitor



WO2011153509
MDM2 inhibitor intermediate

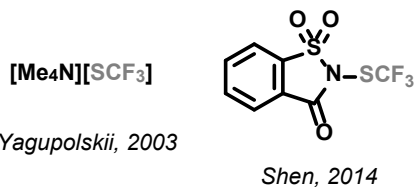


WO2016100542
RAS Gtpase inhibitor

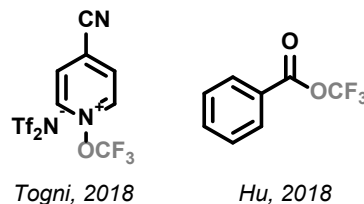
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But how to make them?

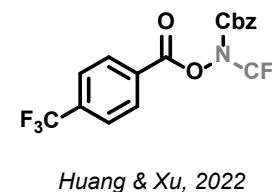
Trifluoromethylthiolation



Trifluoromethoxylation



Trifluoromethylamination



Proposed change in EU law

Potential ban on polyfluorinated alkyl substances (PFAS) in EU?



ANNEX XV RESTRICTION REPORT

PROPOSAL FOR A RESTRICTION

SUBSTANCE NAME(S): Per- and polyfluoroalkyl substances (PFASs)

Proposed change in EU law

Potential ban on polyfluorinated alkyl substances (PFAS) in EU?



ANNEX XV RESTRICTION REPORT

PROPOSAL FOR A RESTRICTION

SUBSTANCE NAME(S): Per- and polyfluoroalkyl substances (PFASs)



Exceptions are anticipated for pharmaceuticals and agrochemicals

Proposed change in EU law

Potential ban on polyfluorinated alkyl substances (PFAS) in EU?



ANNEX XV RESTRICTION REPORT

PROPOSAL FOR A RESTRICTION

SUBSTANCE NAME(S): Per- and polyfluoroalkyl substances (PFASs)



Exceptions are anticipated for pharmaceuticals and agrochemicals

BUT

The ban could still involve reagents or even API intermediates...

Synthesis of CF₃X moieties

We need new synthetic methods that are:

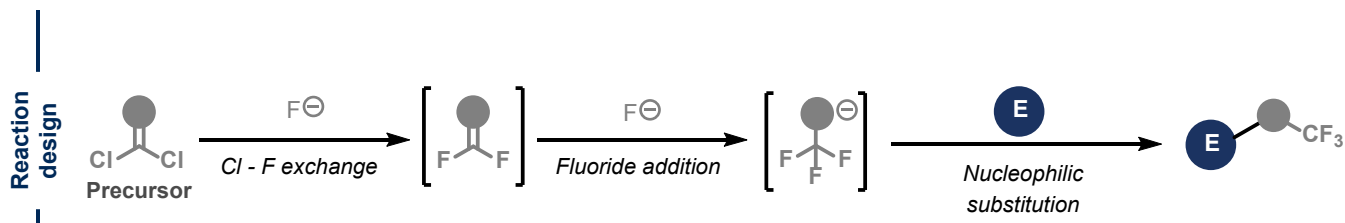
- *Environmentally-friendly*
- *Starting from non-banned chemicals, e.g. alkali fluorides*
 - *Enable Late-stage Functionalization*

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Our approach:

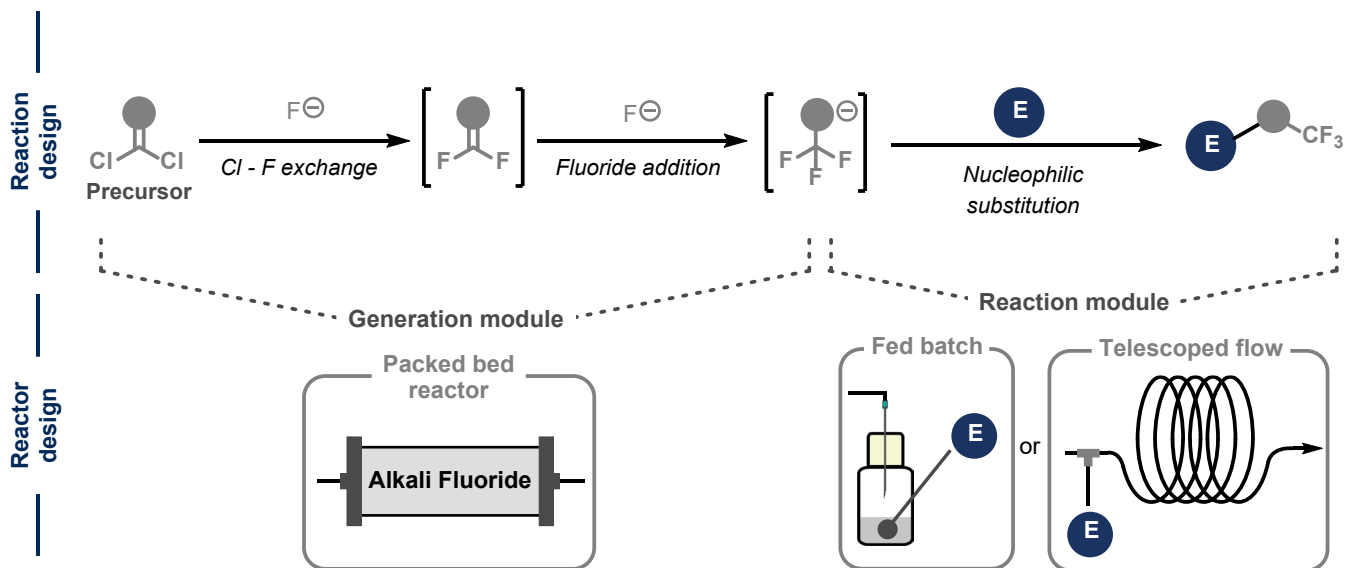


Synthesis of CF_3X moieties

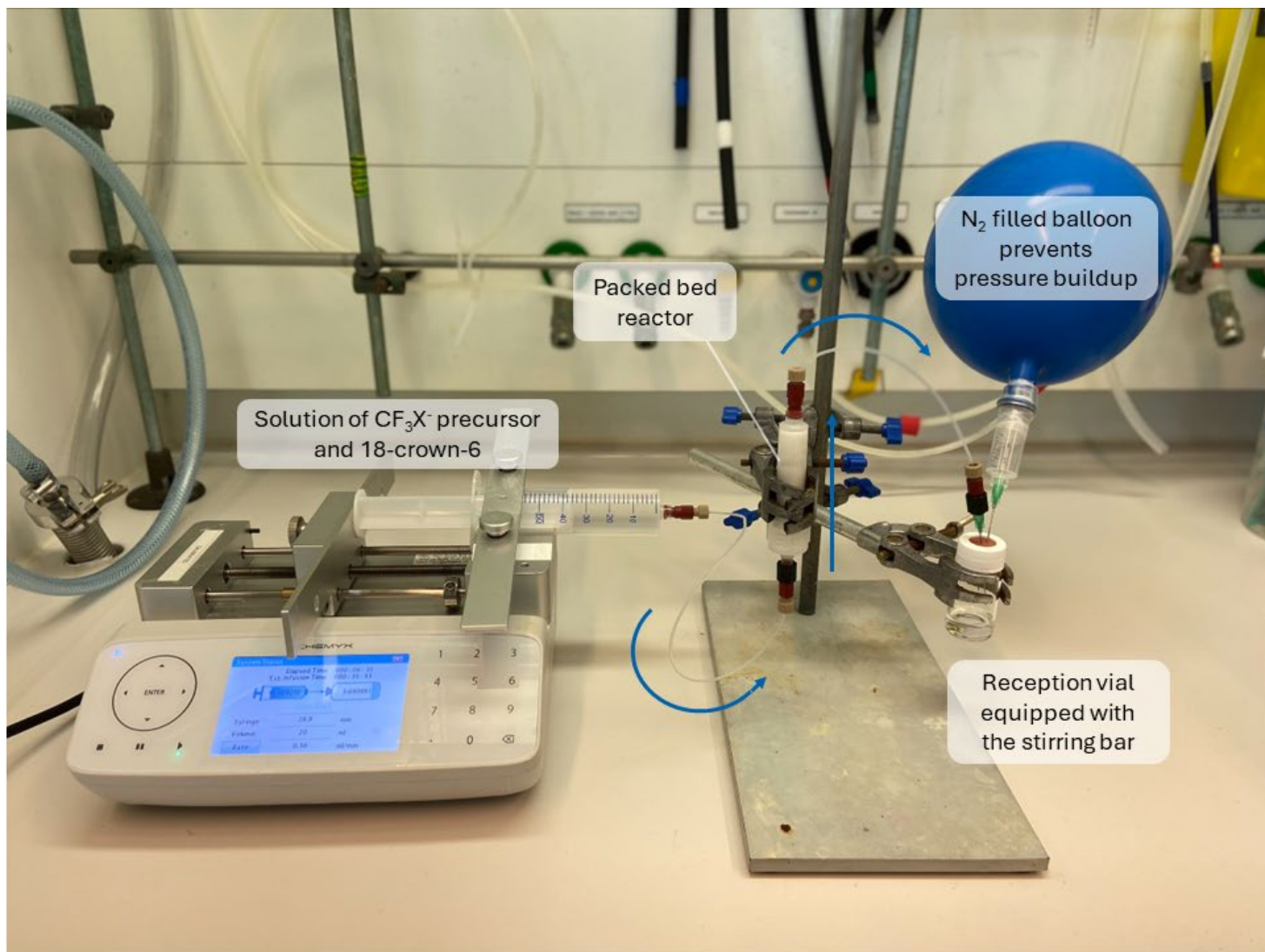
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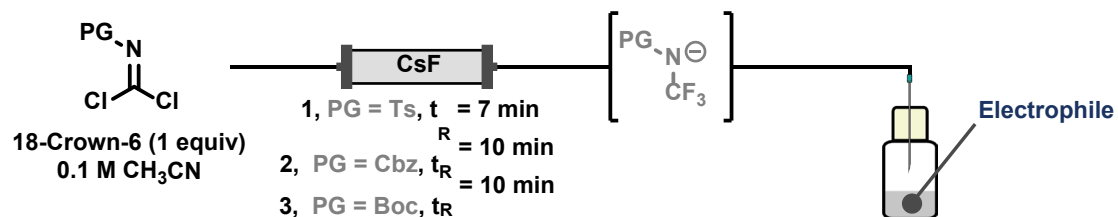
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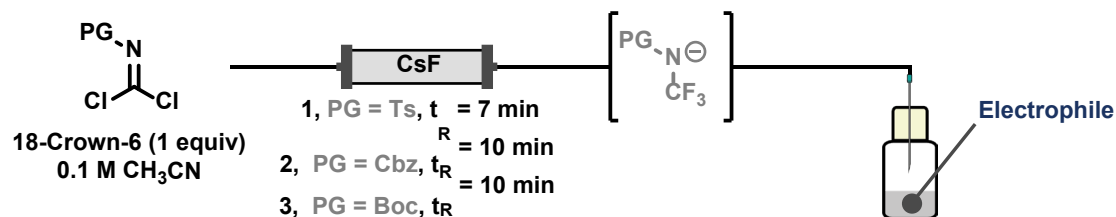
Synthesis of CF_3X moieties



Synthesis of CF₃N anions on demand

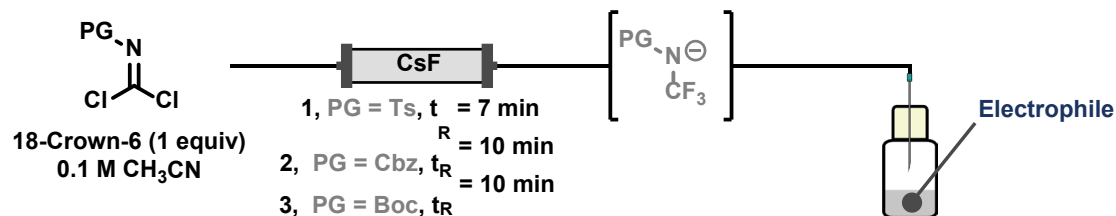


Synthesis of CF₃N anions on demand

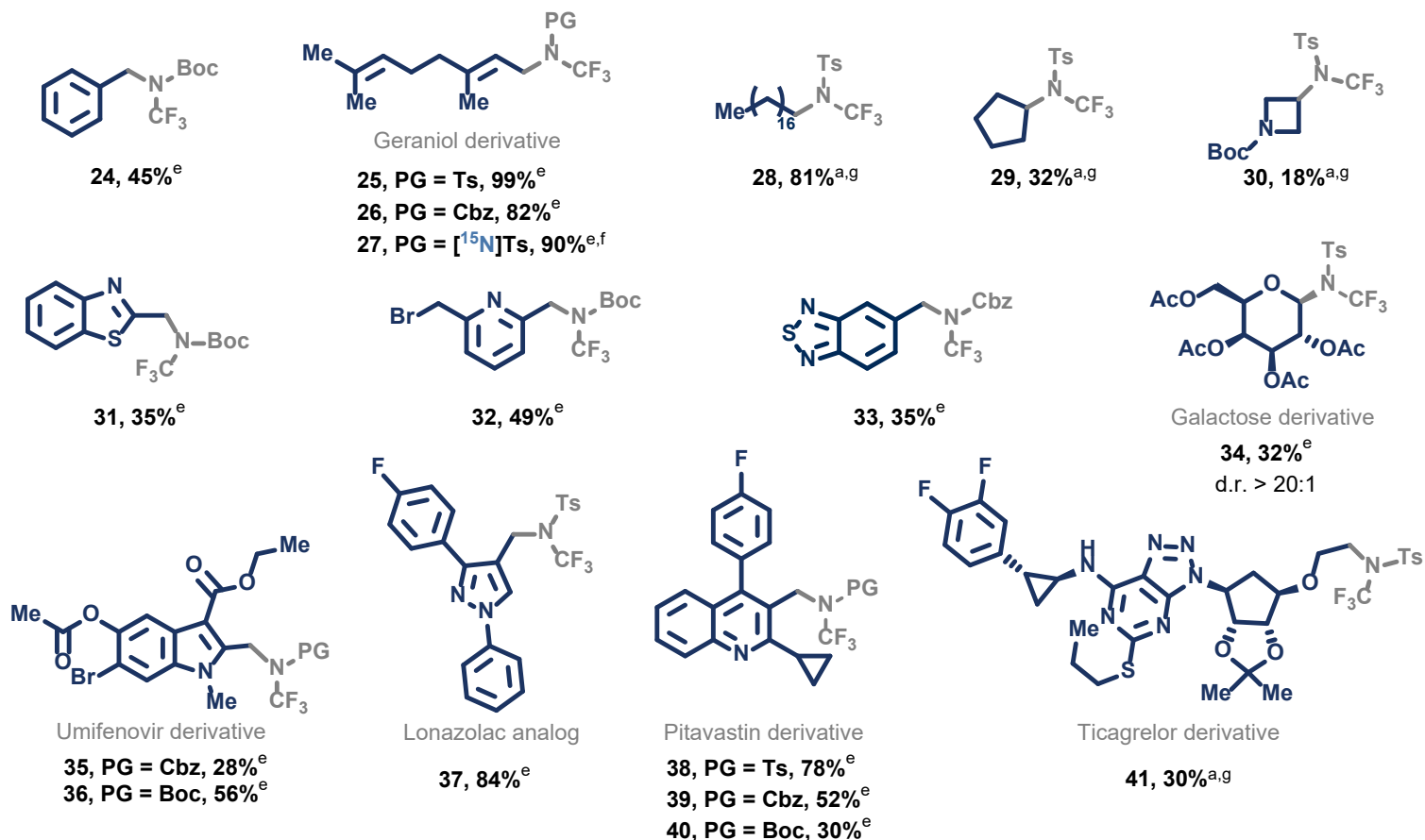


Leaving group assessment Using [CF ₃ TsN] ⁻		Functional group tolerability Using [CF ₃ TsN]						
 4	Br 77% I 98% OMs 59% OTs 79%	 6, 80%	 7, 80%	 8, 79%	 9, 45%	 10, 67%	 11, 62%	
	Br 20% ^a I 99% ^a OMs 7% OTs 19%							 12, 72%
 5	Br 20% ^a I 99% ^a OMs 7% OTs 19%	 18, 37% ^c	 19, 75%	 20, 87% ^d	 21, 61%	 22, 84%	 23, 83% ^d	
								Unsuccessful LGs: Cl, OAc, OTFA, OP(O)OEt, 2,4,6-triphenylpyridinium

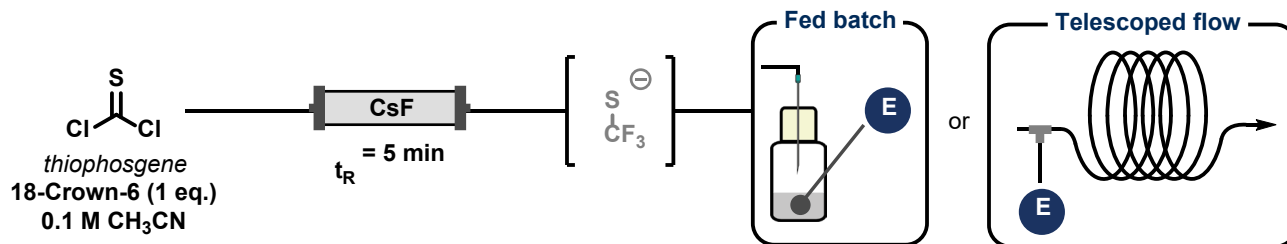
Synthesis of CF₃N anions on demand



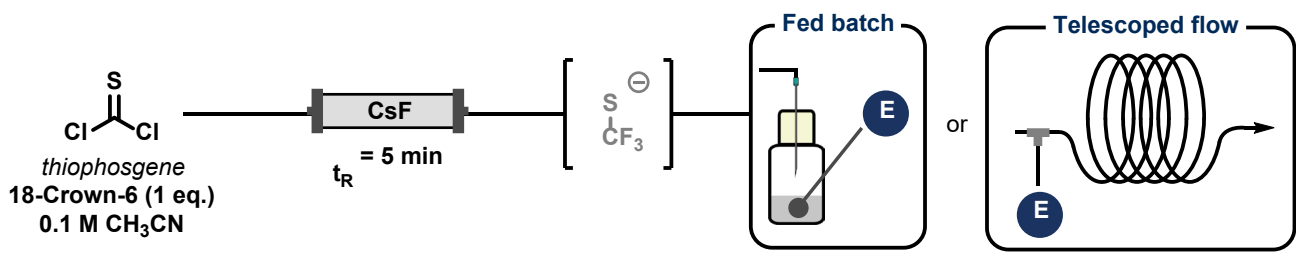
Scope



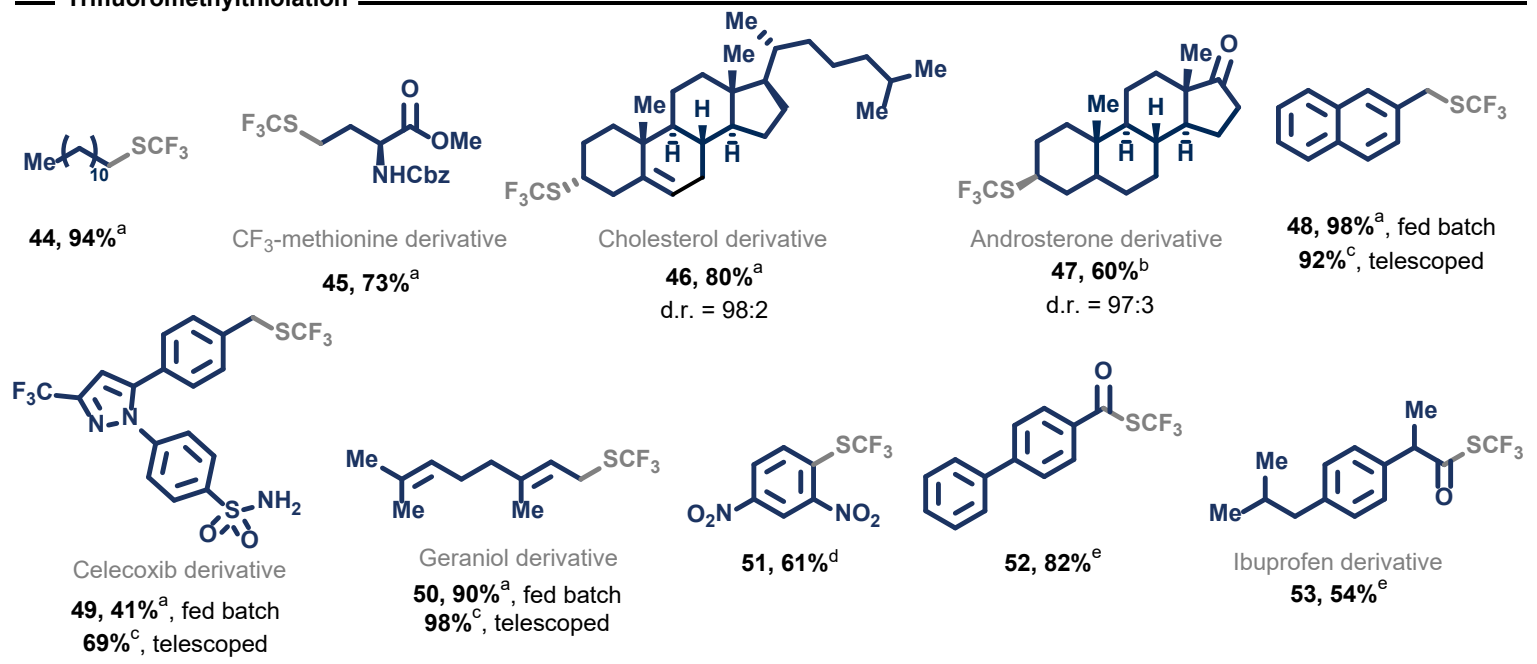
Synthesis of CF_3S^- anions on demand



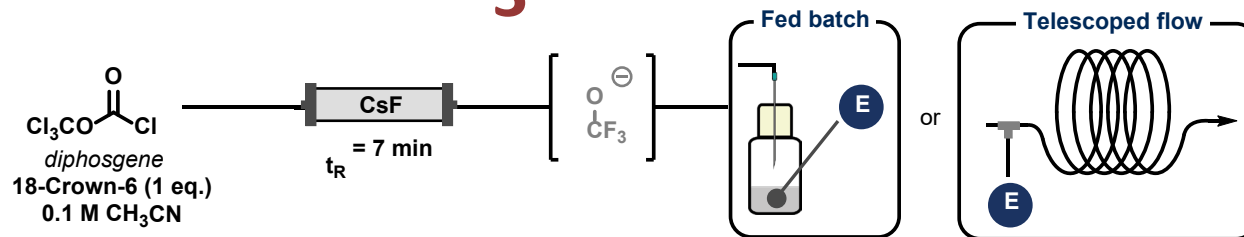
Synthesis of CF₃S anions on demand



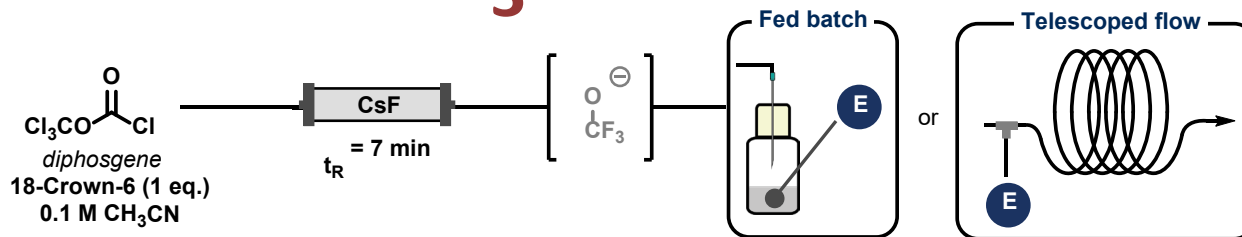
Trifluoromethylthiolation



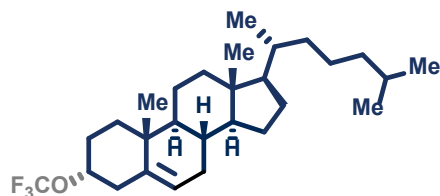
Synthesis of CF_3O^- anions on demand



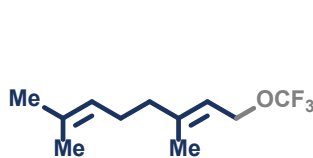
Synthesis of CF₃O anions on demand



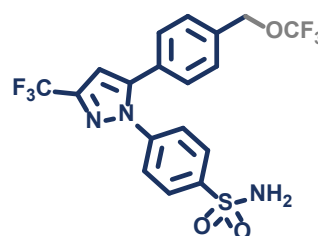
Trifluoromethoxylation



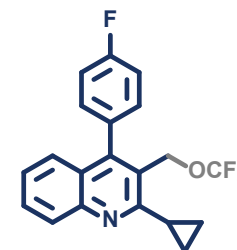
Cholesterol derivative
54, 62%^a
d.r. = 99:1



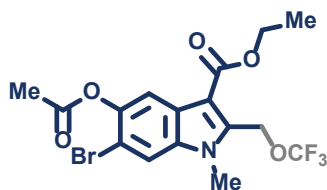
Geraniol derivative
55, 57%^a



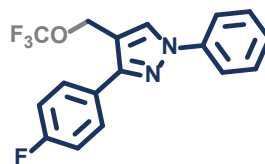
Celecoxib derivative
56, 53%^a



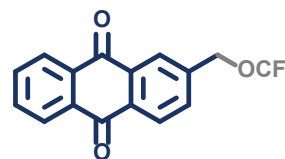
Pitavastatin derivative
57, 80%^a



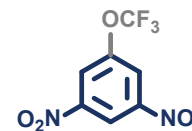
Umifenovir derivative
58, 91%^a



Lonazolac analog
59, 60%^a

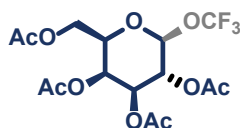


60, 45%^a

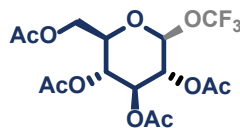


61, 45%^f

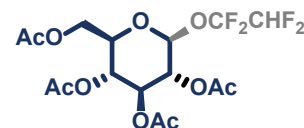
Polyfluoroethoxylation



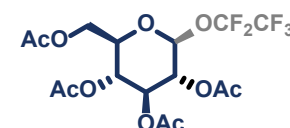
Galactose derivative
62, 99%^a
d.r. > 20:1



Glucose derivative
63, 91%^a
d.r. > 20:1

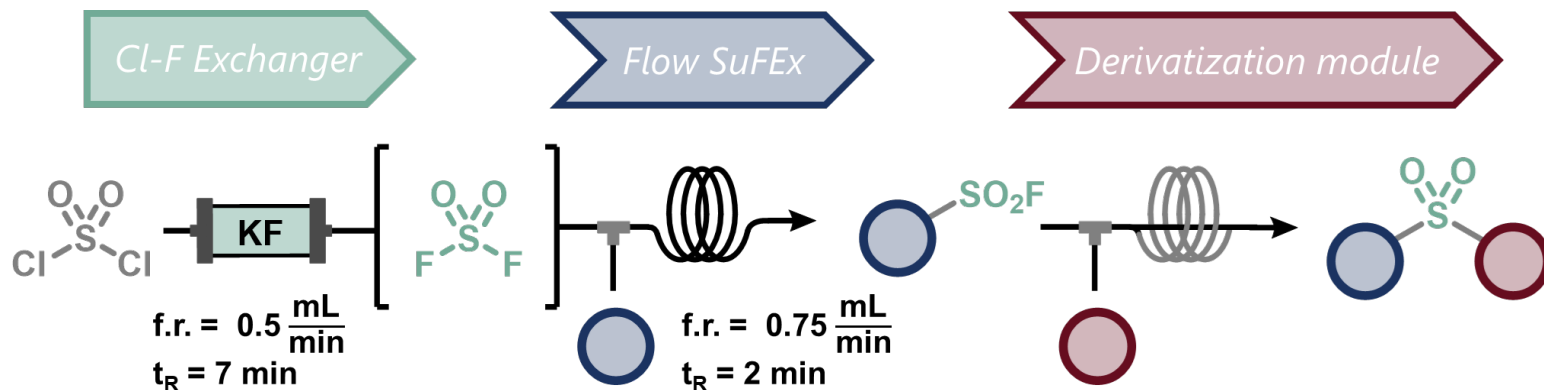


Glucose derivative
64, 18%^{a,g}
d.r. > 20:1

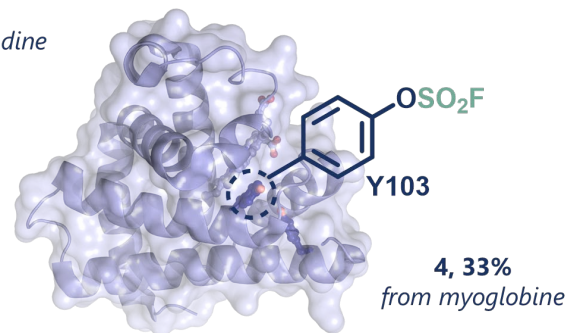
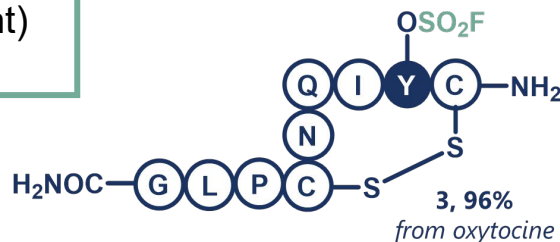
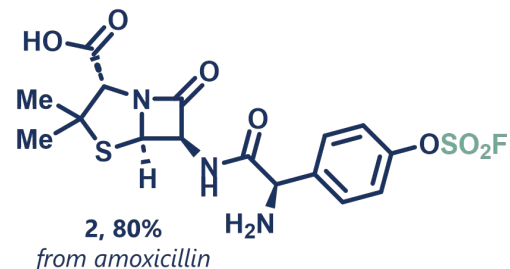
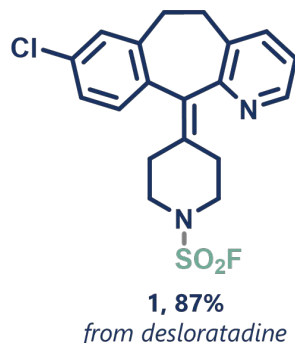


Glucose derivative
65, 37%^{a,h}
d.r. > 20:1

On-demand SO₂F₂ generation

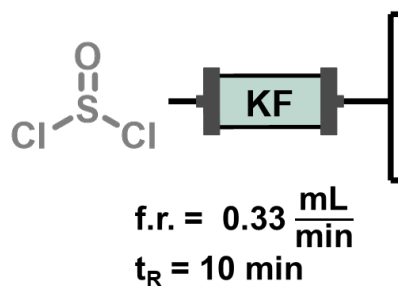


F-S(=O)(=O)-F
sulfonyl fluoride
 difficult to procure
 colorless gas
 toxic (fumigant)

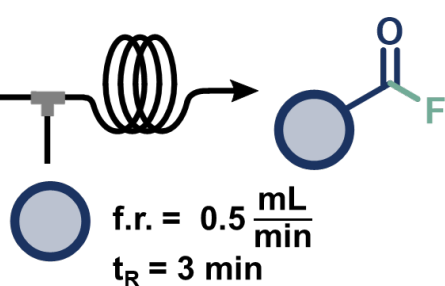


On-demand SOF₂ generation

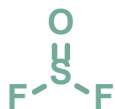
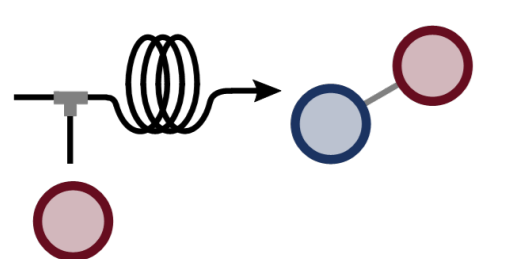
Cl-F Exchanger



Flow deoxyfluorination

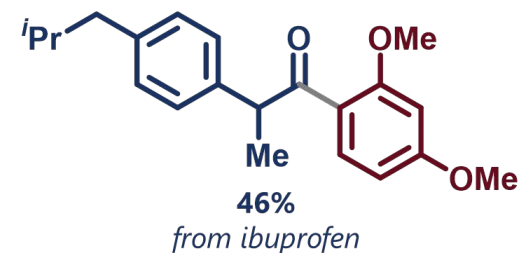
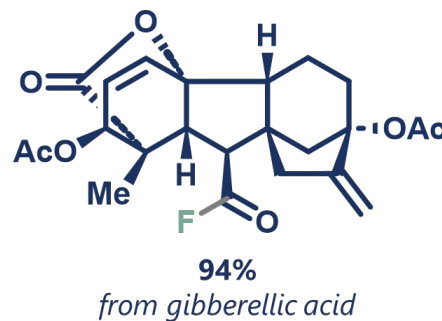
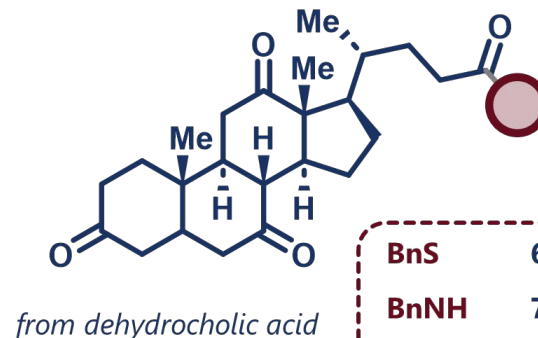
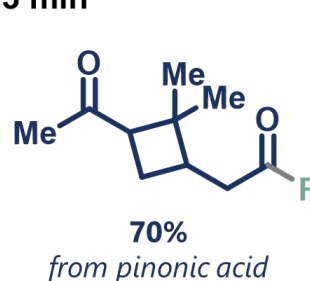


Derivatization module



thionyl fluoride

difficult to procure
colorless gas
toxic (fumigant)
hydrolytic sensitive



Conclusions

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Acknowledgements



Acknowledgements

